Author version: Meteorit. Planet. Sci., vol.46(6); 2011; 857-874

Ion microprobe analyses of oxygen three isotope ratios of chondrules from the Sayh al Uhaymir 290 CH chondrite using a multiple-hole disk

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ABSTRACT

The ion microprobe is the only technique capable to determine high precision stable isotope ratios in individual tiny extraterrestrial particles ($\leq 100~\mu m$ in diameter), but these small samples present special analytical challenges. We produced a new sample holder disk with multiple holes (3 holes and 7 holes), in which epoxy disks containing a single unknown sample and a standard grain are cast and polished. Performance tests for oxygen two isotope analyses using olivine standard grains show that the new multiple-hole disks allow accurate analysis of tiny particles if the particles are located within the 500 μm and 1 mm-radius of the center of holes for 7-hole and 3-hole disks, respectively.

Using the new seven-hole disk, oxygen three isotope ratios of eight magnesian cryptocrystalline (~ 100 μ m in diameter) from the Sayh al Uhaymir (SaU) 290 CH chondrite were analyzed by an ion microprobe at University of Wisconsin. Five out of eight chondrules have nearly identical oxygen isotope ratios ($\Delta^{17}O = -2.2 \pm 0.6$ %; 2SD), which is consistent with those of magnesian cryptocrystalline chondrules in CH/CB and CB chondrites, suggesting a genetic relationship, i.e., formation by a common impact heating event. The other three chondrules have distinct oxygen isotope ratios ($\Delta^{17}O$ values from -6.4 % to +2.2 %). Given that similar variation in $\Delta^{17}O$ values were observed in type I porphyritic chondrules in a CH/CB chondrite, the three chondrules possibly have formed in the solar nebula, similar to the type I porphyritic chondrules.

INTRODUCTION

Studying tiny extraterrestrial particles (≤ 100 µm in diameter), such as Antarctic micrometeorites (AMMs), interplanetary dust particles (IDPs), and cometary particles from the Stardust sample return mission, provide important information about the early evolution of the solar system. Stable isotope anomalies of hydrogen, carbon, nitrogen, and oxygen in these tiny particles detected by ion microprobe (or Secondary Ion Mass Spectrometer, SIMS), which are related to stellar nucleosynthesis and to chemical reactions in molecular clouds (Messenger et al. 2003, 2005; Floss et al. 2006; McKeegan et al. 2006; Yada et al. 2008), provide a clue to the prehistory of the solar system. Some particles from comet 81P/Wild 2 (Stardust particles) are similar to Ca-Al-rich inclusions (CAIs) and chondrules in primitive meteorites in terms of mineralogy, chemistry, and oxygen isotope ratios (Zolensky et al. 2006; McKeegan et al. 2006; Nakamura et al. 2008a), suggesting radial transport of high temperature solids formed in the inner solar nebula to the outer solar nebula region. Thus, stable isotope ratios of tiny extraterrestrial particles serve as a useful tracer for the early solar system evolution.

For effective stable isotope analyses of tiny particles using ion microprobes, multiple samples are placed in a single mount (typically 25 mm diameter) and a maximum polished surface area of individual samples is required. In the case of ion microprobe U-Pb zircon geochronology mounts, more than 100 grains of zircon crystals (typically $\sim 100~\mu m$ in diameter) are mounted together in a single 25 mm epoxy disk, ground to their mid-sections, and polished. However, for Stardust particles that are precious and with smaller sizes (typically less than 10 μm in diameter), it is difficult to obtain maximum polished surface areas for every particle by a single polishing process. It is important to polish the tiny samples embedded with resins individually to obtain maximum polished surface area. Nakamura et al. (2008a) used a custom-made disk with seven holes, in which multiple individually polished epoxy disks are inserted, and performed oxygen three isotope analyses of Stardust particles using an IMS-1280 at University of Wisconsin (WiscSIMS laboratory). For those analyses, a 2 μm diameter Cs⁺ primary beam was used with typical analytical uncertainties of 1 - 2 m (2SD) for m0 and m10 (permil deviations of m180/160 and m170/160 from VSMOW [Vienna Standard Mean Ocean Water]). The 7-hole disk used by

Nakamura et al. (2008a) showed irregular dents around the points that spot-welded a tungsten surface plate and stainless steel disk (the dimension will be described in the next section). The dents may deform electrostatic field on the sample surface. Deformation of electrostatic field modifies the trajectory of secondary ions, which results in isotope fractionation (Kita et al. 2009). In order to avoid any potential problem from the irregular surface and aiming at higher precision and accuracy for analyses, we produced seven-hole disks similar to that used in Nakamura et al. (2008a) (Fig. 1a) and new three-hole disks as modification (Fig. 1b). We performed several tests of oxygen isotope analysis with the multiplehole disks, and assessed the conditions of highly precise and accurate oxygen isotope analysis for tiny particles. Subsequently, oxygen three isotope ratios of tiny chondrules ($\sim 100 \mu m$ in diameter) from the Sayh al Uhaymir (SaU) 290 carbonaceous high-metal (CH) chondrite (Park et al. 2005) were analyzed using the 7-hole disk as an example of the use for unknown samples. The tiny chondrules were extracted previously from the bulk meteorite sample. A fraction of the extracted chondrules were analyzed for noble gases (Nakashima and Nagao 2008). The tiny chondrules are suitable for confirmation of measurement techniques including sample preparation, because they have a similar size to those of AMMs, IDPs and Stardust particles. Furthermore, as described in the next subsection, these chondrules may be related to cryptocrystalline (CC) chondrules in carbonaceous Bencubbin-like (CB) chondrites, which show nearly identical oxygen isotope ratios of $\Delta^{17}O = -2.3 \pm 0.6 \%$ (2SD) (Krot et al. 2010) ($\Delta^{17}O$ = $\delta^{17}O - 0.52 \times \delta^{18}O$). In this respect, we might have selected chondrules with indistinguishable oxygen isotope ratios, which would be useful to evaluate the reproducibility of our analytical protocols on the actual samples.

Chondrules in CH Chondrites

CH chondrites (~ 20 vol% of FeNi metal) share petrologic and isotopic characteristics with more metal-rich CB chondrites (~ 70 vol% FeNi; Weisberg et al. 1995; Krot et al. 2002). In conjunction with the discovery of the Isheyevo CH/CB chondrite which contains two lithologies: metal-rich CB-like and metal-poor CH-like lithologies (Ivanova et al. 2008), a genetic link between CH and CB chondrites has been suggested (Weisberg et al. 1995; Krot et al. 2002; Ivanova et al. 2008). Furthermore, it was

suggested that the CH and CB metal-rich carbonaceous chondrites are related to cometary materials based on the chemical affinities to Stardust particles (Weisberg and Connolly 2008).

Chondrules in CH chondrites are small (~ 20 - 90 μ m in diameter on average) relative to those in other chondrites (0.2 - 5 mm), and 79 % of chondrules in CH chondrites are CC type (including radial pyroxene and glassy chondrules) (Scott and Krot 2003). Such non-porphyritic chondrules formed via complete melting, and their chemistry is considered to have been strongly affected by ambient gas compared to porphyritic chondrules, which formed via incomplete melting. It is expected that the oxygen isotope ratios of CC chondrules can serve as a tracer for the environment of chondrule formation.

CC chondrules in CH and CB chondrites are consistently depleted in volatile lithophile elements including Na and K (Krot et al. 2000, 2001), suggesting that the CC chondrules were isolated from the formation location at elevated temperatures and then cooled rapidly (Krot et al. 2001; Hezel et al. 2003). The abundances of refractory elements such as Ca, Al, and Ti in the CC chondrules are also depleted in variable degrees (Krot et al. 2000, 2001), which is explained by fractional condensation, i.e., more refractory elements condense earlier in the cooling environment (Krot et al. 2001; Hezel et al. 2003; see also Tachibana et al. 2003). In addition, it was reported that magnesian CC chondrules in CH/CB and CB chondrites have similar elemental compositions and show nearly identical oxygen isotope ratios (Krot et al. 2010). Krot et al. (2010) suggested a genetic relationship between the magnesian CC chondrules in the CH/CB and in CB chondrites, i.e., formation by a common heating event induced by large-scale planetesimal impact. Given the chemical affinity between magnesian CC chondrules in CH and CB chondrites (Krot et al. 2000, 2001), one can expect that magnesian CC chondrules in CH chondrites also have similar oxygen isotope ratios to those in magnesian CC chondrules in CH/CB and CB chondrites. Through the mineralogical, chemical, and oxygen isotopic studies of the non-porphyritic chondrules from SaU 290, we discuss the relationship between magnesian CC chondrules in CH and CB chondrites, which facilitates elucidation of the formation environment of chondrules in metal-rich carbonaceous chondrites.

FABRICATION OF MULTIPLE-HOLE DISKS

Seven-hole disks

At the time when Nakamura et al. (2008a) analyzed oxygen three isotope ratios of Stardust particles, the WiscSIMS IMS-1280 could load only two sample holders in vacuum at once (one in the analysis chamber, another in the airlock chamber). For the special requirements of this study, it takes more than a day to reduce the ambient pressure of the analysis chamber containing an epoxy mount in order to reduce ¹⁶OH contribution to ¹⁷O signals, especially at low primary ion intensities, even using a liquid N₂ trap in the analysis chamber. A multiple-hole disk enables a number of samples to be outgassed without exchanging the sample holder. This is the reason why Nakamura et al. (2008a) employed a 7-hole disk for mounting Stardust particles.

The previous 7-hole disk sample holder (25 mm in diameter), which was used by Nakamura et al. (2008a), consists of two parts (Fig. 1c): an upper disk (3.0 mm thick; stainless steel) with a tungsten plate (100 μ m thickness) and a bottom disk (2.0 mm thick; stainless steel). The tungsten plate has seven windows, each with a diameter of 4.0 mm, and is spot-welded on the upper disk that has seven holes each with a diameter of 6.0 mm. The tungsten plate works as a lip to keep the polished sample surface parallel to the disk surface and to make electrical conductivity of the sample surface. Small epoxy disks (~ 5.6 mm in diameter and ≤ 1.7 mm thickness) that contain samples and running standards are put in the upper disk. The bottom disk presses sample surface against the tungsten plate via a spring between sample and the disk, and the bottom disk is screwed to the upper disk (Fig. 1c).

The 7-hole disk used by Nakamura et al. (2008a) showed irregular dents around the welded points on the tungsten plate. In order to avoid any potential problem from the irregular surface (see Introduction), the two new 7-hole disks (disk-1 and disk-2) were fabricated without a welded tungsten-plate (Fig. 1a; disk-1). The new disks have the same dimensions as that of the tungsten-lipped disk, though the upper disks have multiple 4.0 mm diameter windows with a lip machined from a single block of stainless steel. If the surfaces of epoxy disks under the windows are tilted, the measured isotope ratios could be biased (Kita et al. 2009). The flatness of the upper disk surface and epoxy disks was checked

by a ZYGO NewView white light profilometer at the Material Science Center, University of Wisconsin-Madison. The flatness of the entire disk is better than 5 μ m and the tilt of individual epoxy disk is less than ~ 10 μ m across the 4.0 mm window. During the fabrication of the disk-1, the internal edges of the lips were made tapered (~ 70 μ m / 1.0 mm) and slope down to the holes, which is expected to reduce deformation of electrostatic field around the lip. The surface of the lip of the disk-2 was flat and consistently 100 μ m thick.

Three-hole disks

For Stardust particle studies, epoxy (or acrylic) cylinders with a diameter of \sim 8 mm are frequently used (personal communication by M. E. Zolensky). Therefore, multiple-hole disks for the 8 mm epoxy disks were needed. Larger diameter holes reduce total numbers of particles that can be placed in a single holder and require frequent holder exchanges, which may have a disadvantage for achieving ultra-high vacuum (<10⁻⁹ torr) during the analyses. However, in 2009, we installed a new storage chamber to the IMS-1280 that can hold up to 6 sample holders. Therefore, it is not so critical to maximize a number of sample disks to be placed in a single sample holder for achieving desirable high vacuum for the oxygen three isotope analyses.

We produced two 3-hole disks (disk-3 and disk-4) with a diameter of 25 mm (Fig. 1b; disk-3). General design of the 3-hole disks is similar to new 7-hole disks, consisting of upper and bottom disks, both of which are machined from single stainless-steel blocks. Compared to the 7-hole disks, a thickness of the upper disk is increased to 4.0 mm so that epoxy disks with maximum thickness of 2.5 mm can be mounted. The three holes in the upper disks have a diameter of 8.0 mm with a 6.0 mm diameter window and a 100 µm lip. The centers of the three holes are located at 5 mm from the center of the whole disk, which is 5 mm away from the outer lip of the CAMECA holder (Fig. 1b) that has another 100 µm thickness. In the case of the 7-hole disks, the centers of outer six holes are located 7.5 mm from the center of the whole disk, which is much closer to the edge of the outer lip (2.5 mm). Wide windows and proximity of hole centers to the disk center are expected to reduce deformation of the surface electrostatic field. Thickness of the bottom disks and mounting procedure of epoxy disks are the same as

those for 7-hole disks (Fig. 1c).

SAMPLE PREPARATION

Standard olivine grains for test analyses

Epoxy disk preparation procedures are similar to those of Nakamura et al. (2008a). A rod made of epoxy resin (5.6 mm or 7.8 mm in diameter and ~ 10 - 15 mm in length) that has a mesa with a 1 × 1 mm square top is prepared (Fig. 1d). A particle to be analyzed and a running standard are placed on the top, and fixed with a drop of epoxy resin. The mesa is then covered with a plastic tube and filled with epoxy. The epoxy rod is ground and polished using diamond lapping film until the maximum surface area of the particle is exposed. Thus, the particle and running standard are precisely located within the 1 × 1 mm square on the center of the epoxy disk, which is important for isotope analyses using multiple-hole disks. Finally, the bottom of the rod is cut off at ~ 1.5 - 1.7 mm for 5.6 mm epoxy disks (or 2.0 - 2.5 mm for 7.8 mm epoxy disk) from the polished surface.

Chondrule samples

More than three hundred tiny spherical chondrules (100 - 200 μm in diameter; Fig. 2) were previously extracted from the bulk sample of SaU 290 (~ 1.5 g) by a freeze-thaw cycling (~ 50 times) and crushing in a cylinder mortar for noble gas analyses of individual chondrule (Nakashima and Nagao 2008). Eight chondrules (Ch01 - Ch09; Ch03 is unused) were selected randomly for this study from the extracted chondrules.

We followed sample preparation and observation processes of Nakamura et al. (2008a). After the chondrules were examined by synchrotron radiation X-ray diffraction (SR-XRD), all eight chondrules were individually embedded in epoxy disks with San Carlos olivine (SC-OI) running standard grains for oxygen isotope analysis, and the maximum polished surface area was obtained for chondrules. In two epoxy disks, the SC-OI grains were not exposed on the surface (Ch04 and 09). The SC-OI grain in the center hole mounted in the 7-hole disk was used as the running standard for these two chondrule analyses.

ANALYTICAL METHOD

Synchrotron Radiation X-Ray Diffraction

Bulk mineralogy of five chondrules was investigated by SR-XRD at the Institute of Material Structure Science, High Energy Accelerator Organization (Tsukuba, Japan). The chondrules were mounted on a thin glass fiber (5 μ m in diameter) and exposed to monochromatic X-rays (λ = 2.166 and 2.164 Å; two sessions). Clear X-ray powder photographs were taken using a Gandolfi camera with an exposure to synchrotron X-rays of less than 30 min (Nakamura et al. 2008b).

Electron Microscopy

The backscattered electron (BSE) images of the polished surface of chondrules were obtained using a scanning electron microscope (SEM; Hitachi S3400) with an energy-dispersive X-ray spectrometer (EDS) at University of Wisconsin-Madison. Elemental compositions of the chondrules were measured with an electron probe microanalyzer (EPMA; CAMECA SX-51) equipped with five wavelength-dispersive X-ray spectrometers (WDSs) at University of Wisconsin-Madison. WDS quantitative chemical analyses of individual silicate phases were performed at 15 kV accelerating voltage and 6 - 20 nA beam current with a focused beam of \sim 1 μ m in diameter. Analyses of bulk chondrule chemical compositions were performed using 15 kV accelerating voltage and 10 nA beam current with a defocused beam of \sim 25 μ m in diameter.

Oxygen isotope Analyses

We applied three different conditions for oxygen isotope analyses with the CAMECA IMS-1280 ion microprobe at the WiscSIMS Lab: 1) the routine oxygen two isotope analysis ($^{18}O/^{16}O$) using 10 μ m spots (Kita et al. 2009) for multiple-hole disk tests, 2) oxygen three isotope analysis with 15 μ m spots (Kita et al. 2010) for chondrules from SaU 290, and 3) oxygen three isotope analysis with \sim 4 μ m spots for selected chondrule samples. The condition 3) is similar to that in Nakamura et al. (2008a) for Stardust sample analyses, though we used slightly larger beam size and reduced the analysis time per single spot. After SIMS analysis, all SIMS pits were observed with the SEM at University of Wisconsin-Madison to confirm the analyzed positions.

Conditions of analyses

For the test analyses of the multiple-hole disks, a focused Cs⁺ primary beam was set to a size of \sim 10 µm diameter at the intensity of \sim 2.0 nA, and was accelerated by a 20 kV total potential (10 kV in a primary column and -10 kV on the sample). A normal incidence electron gun was used for charge neutralization. Secondary ions of $^{16}\text{O}^{-}$ and $^{18}\text{O}^{-}$ were accelerated at -10 kV to the detectors with high secondary ion transmission. Secondary ion optics was configured similarly to those by Kita et al. (2009): transfer lens magnification of \times 200, contrast aperture 400 µm in diameter, field aperture 4000 \times 4000 µm, entrance slit width 120 µm, energy slit width 40 eV, and exit slit width 500 µm (mass resolving power \sim 2000). Multi-collection Faraday Cups (FCs) were used to measure $^{16}\text{O}^{-}$ and $^{18}\text{O}^{-}$ simultaneously. Typical count rate of $^{16}\text{O}^{-}$ was 3 \times 10 9 cps (converted from ion current). The baseline of the FC amplifiers was measured once before starting the analysis session.

For the oxygen three isotope analyses of the SaU 290 chondrules, two sizes of focused Cs⁺ primary beams ($10 \times 15 \, \mu m$ at the intensity of ~ 2.5 nA and 3 × 4 μm at ~ 28 pA) were applied. Other conditions are generally similar to oxygen two isotope analyses, otherwise described below. The entrance slit width was set to 75 μm . The secondary ¹⁷O⁻ ions were detected by using mono-collector at the ion optical axis and fixed position. The exit slit width for mono-collector detecting ¹⁷O⁻ ions was set to 240 μm (mass resolving power ~ 5000) and 200 μm (mass resolving power ~ 6000) in the sessions with 15 μm spots and with 4 μm spots, respectively, which were sufficient to clearly separate ¹⁷O and ¹⁶O¹H peaks. For detection of minor isotopes ¹⁷O⁻ and ¹⁸O⁻, FC detectors were used in the session with 15 μm spots and electron multipliers (EM) in the session with 4 μm spots. In the session with 4 μm spots, the field aperture was set to 3000 × 3000 μm . In these conditions, intensities of ¹⁶O were 2.6 × 10⁹ and 1.6 × 10⁷ cps with 15 μm and 4 μm primary beams, respectively. The baselines of FCs were measured during the presputtering (100 sec for 15 μm spots and 360 sec for 4 μm spots) in respective analyses, and used for data correction. The contribution of ¹⁶O¹H⁻ tail interference to the ¹⁷O⁻ signal was corrected by the method described in Heck et al. (2010), though the contribution was negligibly small ($\leq 0.1 \, \%$).

Instrumental bias corrections and precision of analyses

For correction of instrumental mass fractionation (instrumental bias), San Carlos olivine (SC-Ol; $\delta^{18}O = 5.32$ % VSMOW, Kita et al. 2010), three kinds of low-Ca pyroxene (En97, En90, and En85), and diopside were used for the session with 15 μ m spots, while for the session with 4 μ m spots SC-Ol, low-Ca pyroxene with En97, and diopside were used.

Actual chondrules also contain minor olivine, glass and silica phases, which overlap within a beam of 15 μ m spot analyses. We corrected instrumental bias assuming as low-Ca pyroxene, except for a silica-bearing chondrule, because these minor phases have small volume and instrumental bias similar to that of low-Ca pyroxene (within a few ‰, Valley and Kita 2009). In the 4 μ m spot analyses of pyroxene, instrumental biases were corrected for individual pyroxene using their respective Wo contents (by linear interpolation between enstatite and diopside standards, Kita et al. 2010). For a chondrule containing silica (for both 15 μ m and 4 μ m spot analyses), instrumental bias was corrected as a mixture of low-Ca pyroxene and silica by estimating their relative fractions from the bulk chemical composition of the chondrule. The instrumental bias correction factor of -9 ‰ relative to SC-Ol is assumed (Valley and Kita 2009). These instrumental bias corrections are mass-dependent, thus do not affect Δ^{17} O values.

In the session with 15 μ m spots, three spots were analyzed in individual chondrules, bracketed by six to eight sets of analyses on the SC-Ol grains mounted with the chondrules, while five to eight spots were analyzed in the chondrules in the session with 4 μ m spots, bracketed by six sets of analyses on the SC-Ol grains. In the two samples where the SC-Ol grain was not exposed on the polished epoxy surface, the SC-Ol grain in the center hole was used as the running standard for the chondrule analyses. External reproducibility of 15 μ m spot analyses of the SC-Ol grains was 0.2 - 1.8 % for δ^{18} O, 0.5 - 1.6 % for δ^{17} O, and 0.5 - 0.9 % for Δ^{17} O (2SD), while that of 4 μ m spot analyses was 0.7 - 1.8 % for δ^{18} O, 0.7 - 1.6 % for δ^{17} O, and 0.7 - 1.1 % for Δ^{17} O (2SD). In one sample, SC-Ol standard mounted in the epoxy disk was relatively smaller (~ 70 μ m) than other mounts and the external reproducibility was poor (1.8 % and 1.6 % for δ^{18} O and δ^{17} O, 2SD, respectively), though it did not affect the reproducibility of Δ^{17} O data significantly.

PERFORMANCE TEST OF MULTIPLE-HOLE DISKS

As test analyses of the multiple-hole disks, we measured oxygen two isotope ratios of SC-Ol standard grains mounted in epoxy disks at WiscSIMS. The reproducibility of oxygen isotope ratios among different holes and within the individual holes in the multiple-hole disks were used to evaluate the performance of the disks.

Test-1: Inter-hole instrumental mass biases relative to the center hole

For the test session of the 7-hole disk-1, four epoxy disks which contain SC-Ol grains (~ 0.1 - 1 mm in diameter) were loaded in the center, west, northwest, and northeast holes (Fig. 1a), and rest of the holes were filled by quartz glass disks with a diameter of 5.6 mm. Three sets of analyses were made on the center of the SC-Ol grains in the outer holes, bracketed by eight sets of analyses (± 0.3 ‰, 2SD) at the center of the SC-Ol grain in the center hole before and after. Next, disk-1 was rotated 180 degrees within the CAMECA holder so that east, southeast, and southwest holes were analyzed. Like the first sub-session, three sets of analyses in the outer holes were made, bracketed by eight sets of analyses in the center hole. Variation of δ^{18} O in SC-Ol grains in outer holes relative to the SC-Ol grain in the center hole is within ± 1 ‰ (Fig. 3a; see also Table 1), which is comparable to the results of test analyses using the tungsten-lipped disk (Nakamura et al. 2008a). Secondary ion yields (16 O count rate / primary beam current; units are 10^9 cps/nA) were comparable [94 - 102 %; (1.27 - 1.38) × 10^9 cps/nA] to those at the center (1.35 × 10^9 cps/nA on average).

For the test session of the 7-hole disk-2, seven epoxy disks containing SC-Ol grains (0.5 - 2 mm in diameter) were loaded, filling all the holes. Four sets of analyses were made on the center of the grains in the outer holes, bracketed by eight sets of analyses (\pm 0.22 ‰, 2SD) at the center of the grain in the center hole before and after. Variation of δ^{18} O in SC-Ol grains in outer holes relative to the SC-Ol grain in the center hole is within \pm 0.5 ‰ (Fig. 3b; see also Table 1), which is better than the disk-1. Secondary ion yields were comparable (92 - 101 %) to those at the center. Since deformation of electrostatic field should reflect the geometries around the respective holes including the outer lip of the

CAMECA holder, isotopic fractionation is expected to be related to the hole position within the disks (X-Y effect; cf., Kita et al. 2009). However, comparing the test results of disk-1 and -2 (Fig. 3), the δ^{18} O variation does not appear to be related to hole positions, i.e., no detectable geometric effect on δ^{18} O in the 7-hole disks overall.

For the test session of the 3-hole disks (disk-3 and disk-4), six epoxy disks (\sim 7.8 mm diameter) containing SC-Ol grains (1 - 4 mm in diameter) were loaded in all of the holes. Four sets of analyses of δ^{18} O were made on the center of the grains in the southeast and southwest holes, bracketed by eight sets of analyses (\pm 0.14 % for disk-3 and \pm 0.21 % for disk-4, 2SD) at the center of the grain in the north hole before and after. The δ^{18} O variation of SC-Ol grains in southeast and southwest holes relative to the SC-Ol grain in the north hole is within \pm 0.5 - 0.7 % (Fig. 4; see also Table 2). Secondary ion yields were comparable (93 - 100 %) to those at the center. Like the 7-hole disks, no geometric effect on δ^{18} O in the disks overall was observed.

Test-2: Intra-hole instrumental mass biases

We observed $\delta^{18}O$ variation at different locations (north, south, east, and west) in each hole. In the two large SC-Ol grains ($\sim 0.5 \times 1$ mm; center and west; first sub-session) in the disk-1, $\delta^{18}O$ variations at different locations relative to the center of the grains were ± 0.5 % (Online supplements). Secondary ion yields were 97 - 103 % relative to those of center of the grains. In disk-2 (all the holes), $\delta^{18}O$ variations were ± 0.5 % within a ~ 500 µm-radius from the center of respective holes (Online supplements). Secondary ion yields were 92 - 104 % relative to those of center of the grains. If analyzing further from the center of the mount (up to 1 mm-radius), relative $\delta^{18}O$ values sometimes exceeded the ± 1 % range, and secondary ion yields decreased to 79 %. These results show that oxygen isotope analyses performed within a ~ 500 µm-radius of the center of each hole do not have significant analytical bias. Using flat quartz glass disks, it was found that the secondary ion yields decrease by ~ 5 % outside of 1 mm-radius. This indicates that the electrostatic field may be more deformed towards the lip of the window, which may lead to deviations of the secondary ion trajectory even though automatic centering of the secondary ions to the field aperture is applied (Kita et al. 2009).

For the 3-hole disks, δ^{18} O variations at different locations (north, south, east, and west) in respective holes were within \pm 0.5 ‰ when measuring within a \sim 1 mm-radius from the center of respective holes, and secondary ion yields were 91 - 102 % relative to those of center of the grains (Online supplements). Outside of the 1 mm-radius, relative δ^{18} O values exceeded the 1 ‰ range, though relative secondary ion yields were 91 - 98 %. Thus, oxygen isotope analyses performed within the 1 mm-radius of the center of each hole may not show significant analytical bias due to location of sample.

Evaluation of multiple-hole disks

The test analyses show that the new multiple-hole disks are applicable to tiny particle analysis of $\delta^{18}O$ with precision of \pm 0.5 % (2SD) as long as the particles are located within the radius of \sim 500 μ m and 1 mm of the center of holes for 7-hole disks and 3-hole disks, respectively. The requirement for particle location is met by the sample preparation procedures described above (see Sample preparation). It is important to mount running standards with samples in the same epoxy disks so as to correct unexpected instrumental fractionation (see the next section). The level of reproducibility at 0.5 % is somewhat worse than that typically obtained from flat 25 mm mounts without significant topography (\pm 0.3 %; Kita et al. 2009). Therefore, the use of multiple-hole sample mounts creates a trade-off of precision and accuracy vs. analysis of a single flat 25 mm mount. However, the analytical uncertainty of $\delta^{18}O \pm 0.5$ % is still smaller than that for analyses with small spots in this study and for Stardust particles (Nakamura et al. 2008a) at the level of 0.7 - 2 %, and does not become a problem for the small spot analyses.

For data obtained from outside of 0.5 ‰ area, we observed systematic change in δ^{18} O values that depended on the location of analyses relative to the center of each hole (Fig. 5). As shown in Fig. 5a (7-hole disks), the relative δ^{18} O values are high in north and east positions, while they are low in the south and west. The 3-hole disks show distribution of relative δ^{18} O values similar to that for the 7-hole disks (Fig. 5b), though δ^{18} O variation is less pronounced for the 3-hole disks. Thus, unlike comparison of relative δ^{18} O values among holes, there is a relationship between relative δ^{18} O values at different locations in each hole and directions, which is so called the X-Y effect (Kita et al. 2009). However, the

trend we observed is opposite to that in Kita et al. (2009) who measured grains with surface topography due to polishing relief in a single flat 25 mm mount. Slightly convex surfaces of individual grains are common in zircon geochronology and other grain mounts (Kita et al. 2009). This contrasts with the disks with multiple holes, the sample surface is 100 μ m deeper than the lip of the disk (i.e., small dimples vs. small bumps). The deformation of the electrostatic field above the sample surface in the multiple-hole disks would thus be in the opposite direction compared to that of zircon grains with polishing relief. Thus, the X-Y effects we observe likely result from the specific geometry of the multiple-hole disks. The 3-hole disks have wider windows and holes closer to the disk center, which reduce deformation of electrostatic field on the sample surface. As a result, the δ^{18} O variations in the 3-hole disks are smaller than that of 7-hole disks.

ANALYSES OF THE SAU 290 CHONDRULES

Mineralogy and chemistry of the SaU 290 chondrules

We studied 8 cryptocrystalline (CC) chondrules. All eight chondrules have spherical shape and their sizes range from 60 µm to 160 µm in diameter. They show various colors under an optical microscope before mounting: white-clouded/transparent and black opaque (Fig. 2; Table 3). We recognized that the chondrules mainly consist of low-Ca pyroxene by SR-XRD (Fig. 8). All crystalline chondrules exhibit fine-grained texture with BSE images (Figs. 6-7), indicating that they are CC chondrules. Three opaque chondrules (Ch01, 02, and 07) contain tiny Fe metal grains (< 1 µm in diameter; detected by SEM-EDS; Figs. 6-7), while others are free from Fe metal grains. The opacity may be due to the presence of the numerous tiny Fe metal grains (Read 1986). Except for the relationship between tiny Fe metal grains and opacity, there appears to be no relationship between the optical properties and texture (or chemistry) for CC chondrules (discussed in the next section). It is not surprising to find all eight chondrules are CC type despite random selection, because ~ 80 % of the chondrules in CH chondrites are CC type (Scott and Krot 2003). In addition, the initial selection of ~ 300 chondrules are biased to only spherical chondrules in order to avoid potential matrix materials (Nakashima and Nagao 2008), which biased our selection towards non-porphyritic chondrules and away from porphyritic chondrules having irregular

shapes (e.g., Wasson 1996).

For CC chondrules, the estimated bulk chemical compositions obtained by defocused-EPMA are close to stoichiometric magnesian low-Ca pyroxene with Mg# (molar % [MgO]/[MgO+FeO]) > 98 (Fig. 9; Table 3). This is consistent with the fact that magnesian CC chondrules are dominant in CH chondrites (e.g., Scott 1988). The bulk chemical compositions of Ch02, 04, 05, 07, and 08 are deviated slightly from stoichiometric low-Ca pyroxene (Fig. 9). The SR-XRD patterns of Ch02, 04, and 05 (no SR-XRD data for Ch07 and Ch08) show reflection lines of olivine in addition to low-Ca pyroxene (Figs. 8a-c). Although we did not recognize olivine crystals by SEM observation, chemical composition data of these chondrules with focused-EPMA appear to be mixture of low-Ca pyroxene and olivine (~ 15 % contribution of olivine), suggesting olivine and low-Ca pyroxene occur as smaller grains than elastic scattering volume of electrons (a few µm in diameter) by EPMA. The slightly high (Mg+Fe)/Si ratios (Fig. 9) are consistent with a small contribution of olivine (11 - 16 %). In Ch02, we identified two phases (Figs. 6b and 7b). The brighter phase has higher Si/Mg atomic ratios (~ 0.87) and Cr₂O₃ content (0.71 wt%) than darker phase (Si/Mg atomic ratios of 0.80 - 0.84, and Cr₂O₃ content of 0.39 wt%; Table 4). In the enlarged BSE image (Fig. 7b), the darker phase shows dendritic texture, suggestive of supercooling (Lofgren 1996). As in the case of Ch02, with a high magnification and enhanced contrast BSE images, we found fine textures in individual CC chondrules (Fig. 7) that can explain the observed variation of the bulk chemical compositions. The bulk chemical composition of Ch01 is slightly shifted towards anorthite from stoichiometric low-Ca pyroxene (Fig. 9), indicating a small contribution of anorthite component (13 % on average). Low-Ca pyroxene occurs as fan-like laths (~ 1 μm in width; ~ En96Wo3; Lpx in Figs. 6a and 7a) with feldspathic glass (~ An56Ab34; Table 4). Note that accuracy of chemical compositions of low-Ca pyroxene and glass by focused-EPMA is limited because of the small grain size of the two phases. In glass, secondary alteration phases such as nepheline or sodalite were not observed, suggesting that Ch01 escaped secondary alteration. High-Ca pyroxene (Hpx, 5 - 15 µm in diameter; En77Wo22; Table 4) occurs with tiny Fe-metal grains (< 1 μm in diameter) on the periphery of Ch01 (Figs. 6a and 7a). The bulk chemical composition of Ch06 is slightly enriched in Si relative to

others. Ch06 shows fine-grained texture (Figs. 6e and 7d), and consists mainly of two types of low-Ca pyroxene (En94Wo5 with 1 - 5 μ m in diameter and En98Wo1 with 7 - 10 μ m in diameter), and SiO₂-rich phase (1 - 2 μ m in diameter; SiO₂ ~ 80 wt%) as shown in Table 4. It is concluded from the SR-XRD pattern that SiO₂-rich phase is cristobalite (Fig. 8d). Focused-EPMA that aimed SiO₂-rich phase detected certain amounts of Al₂O₃, MgO and CaO (Table 4), which may be contaminant from surrounding pyroxene because the size of SiO₂-rich phase is comparable (or smaller) to the scattering volume of electrons by focused-EPMA (a few μ m in diameter). Thus, Ch06 is a silica-bearing chondrule. A contribution of silica to the whole chemical compositions is 14 % on average.

The bulk chemical composition of Ch09 was not obtained with defocused-EPMA due to many voids (Fig. 6h), which results in low oxide total and, unreliable data. Focused-beam EPMA data (avoiding voids) showed chemical composition of Ch09 is close to low-Ca pyroxene (En94Wo5; not shown in Fig. 9; analytical totals ~ 80 - 97 wt%), which is consistent with the SR-XRD pattern for this chondrule that only shows reflection lines of low-Ca pyroxene (Fig. 8e). Thus, Ch09 is also a magnesian pyroxene-dominated CC chondrule.

The CI-normalized elemental abundances of CC chondrules (Fig. 10; Ch09 are not plotted) are within the range of chondrules in other CH chondrites (Krot et al. 2000). The five "typical" CC chondrules (Ch02, 04, 05, 07, and 08) are depleted in refractory and volatile elements, like CC chondrules in other CH chondrites (Fig. 10; Krot et al. 2000). Compared to other typical CC chondrules, Ch01 and 06 are enriched in refractory elements and volatile elements, respectively (Fig. 10). The refractory element abundances in Ch01 exceed those of CI chondrites by the factor of 2, which makes Ch01 to be relatively "refractory-rich" chondrule. Thus, eight chondrules in this study include typical, refractory-rich, silica-bearing CC chondrules.

Oxygen isotope ratios

In the first session (session 1), chondrules were analyzed with 15 µm spots, except for Ch09 that has a porous texture at a scale much smaller than the SIMS beam spots. In the second session (session 2), selected chondrules (Ch01, 02, 04, 06, and 09) were analyzed with 4 µm spots to examine internal

isotope heterogeneity.

At the time of the session 1, chondrules Ch02, 05, 06, 07, and 08 were first loaded to disk-1 at the outer holes for the SIMS analyses, because we only had one 7-hole disk fabricated. After these five chondrules were analyzed, the sample holder was removed from the instrument and the rest of the chondrules Ch01 and 04 were loaded to the same disk for the analysis. As described earlier, a standard olivine grain in the epoxy disk of Ch04 was not exposed to the surface, and therefore standard olivine at the center hole was used as a running standard. The raw measured δ^{18} O values of SC-Ol grains in the center hole and in an outer hole with Ch01 agreed within 0.5 % (Fig. 11), as we expect from the performance test (Fig. 3). However, during the analyses of first set of chondrules, it was observed that raw δ^{18} O values of SC-Ol grains in the outer holes are consistently lower by 1 % than that of the center grain (Fig. 11). The cause of lower δ^{18} O values in outer holes were not identified, though it may be related to sample preparation such as position of epoxy disks within the holes and position of the 7-hole disk within the CAMECA holder. For samples in the first disk, standard oliving grain in each of the outer holes was used to correct instrumental bias. The bias between the central hole and outer holes did not affect any actual data after the instrumental bias corrections. However, this situation would cause systematic error if the standard grain had not been mounted with each unknown chondrule sample that was analyzed during the first set of session 1. Thus, it is very important to mount running standards with samples in the same epoxy disks so as to correct (unexpected) instrumental fractionation. It should be mentioned that raw $\delta^{18}O$ and $\delta^{17}O$ values of standard in the outer disk shift along the slope 0.5 mass fractionation line (Fig. 11). Therefore, results obtained without standard grain in the same epoxy disk have no measurable effect on the Δ^{17} O values.

Oxygen isotope analyses of seven chondrules with 15 μ m spots are shown in Fig. 12a. These data plot along a slope-1 line between the Carbonaceous Chondrite Anhydrous Mineral (CCAM) and Young and Russell (Y&R) lines (Fig. 12). Five chondrules (Ch02, 04, 05, 07, and 08) plot within a narrow range of isotope ratios with the average Δ^{17} O value of -2.2 \pm 0.6 % (2SD; Fig. 13). The other two chondrules have distinct Δ^{17} O values; -6.4 % in Ch06 and +2.2 % in Ch01, as shown in Fig. 13. Data

from three 15 μ m spots in each chondrule are consistent within the analytical uncertainties (Fig. 12a; Table 5). The constituent mineral phases of the chondrules are much smaller than 15 μ m, and therefore the derived oxygen isotope ratios represent average values.

Oxygen isotope analyses of Ch01, 02, 04, and 06 with 4 µm spots are shown in Fig. 12b. In Ch01, 02, and 04, results of 4 µm spot analyses in pyroxene grains with different chemical compositions do not show any resolvable internal isotope heterogeneity beyond analytical uncertainties (Fig. 12b; Table 6). The average δ^{18} O, δ^{17} O, and Δ^{17} O values of Ch01, 02 and 04 are generally in good agreement with those obtained from 15µm spot data. However, oxygen isotope ratios analyzed with 4 µm spots (Fig. 12b) are slightly shifted to the right relative to those analyzed with 15 μm spots (Fig. 12a), i.e., the $\delta^{18}O$ values are slightly higher (+0.9 % on average; Tables 5 and 6). Difference of ~ 1 % between data in δ^{18} O values of two sessions are only marginal compared to the analytical uncertainties (0.7 - 1.8 %) of the second session. The measured δ^{18} O and δ^{17} O values in Ch06 using 4 um spots show a large variation along the slope of ~ 0.5 , which appears to be correlated with the fraction of silica phase as revealed by the examination of the analyses pits after using SEM. The 4 µm primary beam size was comparable to the grain sizes of pyroxene and silica in the chondrule (Fig. 7d). The volume fractions of silica within the pits were difficult to estimate. Thus, data are corrected using the same instrumental biases as that used for 15 µm spot analyses (e.g., average fractions of pyroxene and silica), which do not accurately correct contribution of silica at each analysis point. Therefore, the variations of Ch06 data in Fig. 12b along slope 0.5 are not true variations, but analytical artifacts.

Chondrule Ch09 was only analyzed in the session with 4 μ m spots in order to avoid numerous void spaces within the sample (Fig. 6h). However, SEM observation after the SIMS analyses revealed that all the SIMS pits more or less overlapped the voids. As a result, the 16 O count rates during the analyses were significantly lower (~ 60 %) than those of other chondrules and SC-Ol standard grains, and the internal errors associated with the isotope ratios are about 1.4 times higher than those of the SC-Ol standard grain (Online supplements). The measured δ^{18} O and δ^{17} O values from multiple spot analyses in Ch09 scattered significantly more than those in other chondrules, in the range as large as 10 % in δ^{18} O,

possibly due to the effect of charging by the presence of voids within the analysis areas. However, $\Delta^{17}O$ values of multiple analyses were reproducible within analytical uncertainty with the average value of -6 \pm 2 % (Online supplements). Thus, in spite of poor analytical conditions, Ch09 seems to be another relatively ¹⁶O-rich chondrule similar to Ch06.

FORMATION ENVIRONMENT OF THE SAU 290 CHONDRULES

The average Δ^{17} O value of -2.2 ± 0.6 ‰ (2SD; Fig. 13) observed in the five chondrules (Ch02, 04, 05, 07, and 08) is consistent with those of magnesian CC chondrules in CH/CB and CB chondrites (Krot et al. 2010) and close to those of the most CC chondrules in other CH chondrites (Yoshitake and Yurimoto 2004; Jones et al. 2005; Yurimoto et al. 2008). In addition, the five chondrules are chemically "typical" in terms of depletion in refractory and volatile elements, like CC chondrules in CH and CB chondrites (Fig. 10; Krot et al. 2000, 2001). The identical oxygen isotope ratios and chemical affinity support the view that most of the magnesian CC chondrules in CH and CB chondrites formed together from an isotopically uniform gaseous reservoir generated by an impact event (Krot et al. 2010).

Three other chondrules have distinct $\Delta^{17}O$ values ($\sim +2.2$ % for refractory-rich Ch01, ~ -6.4 % for silica-bearing Ch06, and ~ -6 % for porous Ch09) that obviously could not be formed by the impact heating event common to other magnesian CC chondrules with $\Delta^{17}O$ value of -2.2%. The positive $\Delta^{17}O$ values of $\sim +2.2$ % in Ch01 are rare for chondrules in carbonaceous chondrites, because carbonaceous chondrite chondrules generally show negative $\Delta^{17}O$ values (cf., Krot et al. 2006). Many ordinary chondrite chondrules and a subset of enstatite chondrite chondrules have positive $\Delta^{17}O$ values ($\leq +1.6$ %; Clayton et al. 1991; Kita et al. 2010; Weisberg et al. 2010), though the $\Delta^{17}O$ value of Ch01 is higher than those of chondrules in ordinary and enstatite chondrites and close to those of chondrules in a Rumuruti group (R) chondrite (Greenwood et al. 2000). The chondrules in an R chondrite contain a certain amount of iron (Fa10 and Fs5-7; Greenwood et al. 2000), while Ch01 is almost iron-free (Fs<1). There appears to be no chemical affinity between Ch01 and the chondrules in an R chondrite.

Among carbonaceous chondrite chondrules, chondrules in metal-rich carbonaceous chondrites exceptionally show positive Δ^{17} O values (Yoshitake and Yurimoto 2004; Yurimoto et al. 2008; Krot et al.

2010). Type I porphyritic chondrules in a CH/CB chondrite have Δ^{17} O values from -5 ‰ to +4 ‰ and were suggested to have been a nebular product but not a product of the impact event that formed magnesian CC chondrules (Krot et al. 2010). The range of Δ^{17} O values of Ch01, 06, and 09 (-6 ‰ to +2.2 ‰) are within (similar to) that of the type I porphyritic chondrules (Krot et al. 2010). It may be possible that these chondrules with distinct Δ^{17} O values formed together with the type I porphyritic chondrules in the solar nebula.

The difference in chemical compositions of Ch01, 06, and 09 (Fig. 10) can be explained by fractional condensation in the solar nebula, i.e., more refractory elements condensed earlier in the cooling environment of solar composition and the condensates were isolated from the reactive environment (see Introduction). If the CC chondrules formed by the formation sequence, refractory-rich Ch01 condensed earlier than Ch06 and 09 (Fig. 10 and Table 3; Ca/Si and Al/Si ratios of Ch09 do not exceed CI abundance). Since silica condenses after pyroxene during fractional condensation (Petaev and Wood 1998), silica-bearing Ch06 may have condensed after porous Ch09, which consists only of low-Ca pyroxene. If the three chondrules formed by fractional condensation, they may have formed in the following order: refractory-rich Ch01 (Δ^{17} O ~ +2.2 %), porous Ch09 (~ -6 %), and silica-bearing Ch06 (~ -6.4 %).

The CC chondrules from SaU 290 show various degrees of color and optical transparency (Fig. 2). Based on the mineral assemblage, chemistry, and oxygen isotope ratios, the CC chondrules from SaU 290 are classified into four types: typically with $\Delta^{17}O$ of -2.2 ‰ (black opaque and white-clouded/transparent), refractory-rich with $\Delta^{17}O$ of +2.2 ‰ (black opaque), silica-bearing with $\Delta^{17}O$ of -6.4 ‰ (white-clouded/transparent), and porous with $\Delta^{17}O$ of -6 ‰ (white-clouded/transparent). Chondrules with various colors are included in the typical type, and there appears to be no relationship between color and classification. The typical chondrules have $\Delta^{17}O$ values of -2.2 ‰, while chemically distinct chondrules have distinct $\Delta^{17}O$ values. There may be a relationship between chemistry and oxygen isotope ratios, although statistics are still poor. More data will reveal the relationship between chemistry and oxygen isotope ratios for magnesian CC chondrules.

CH chondrites may be related to Stardust particles (see Introduction; Weisberg and Connolly 2008). Here we compare the SaU 290 chondrules and chondrule-like Stardust particles. The Δ^{17} O values of -2.2 ‰ observed from typical CC chondrules in the SaU 290 CH chondrite are identical to Δ^{17} O values of -2 ‰ observed from a majority of chondrule-like Stardust particles (Nakamura et al. 2008a; 2009). However, the latter shows porphyritic texture and the Δ^{17} O values of -2 ‰ were observed in pyroxene grains with Mg# of 96 - 86 (Nakamura et al. 2008a). The CH chondrite chondrules in this study (Fig. 6 and Table 3) are texturally and chemically different from the chondrule-like Stardust particles, and there seems to be no clear relationship between them.

CONCLUSIONS

We developed new multiple-hole disks (with 7-holes and 3-holes) for high precision stable isotope analyses of tiny particles such as AMMs, IDPs, and Stardust particles using an ion microprobe. Performance tests show that the new multiple-hole disks are applicable to tiny particle analysis as long as the particles are located within the $\sim 500 \mu m$ -radius of the hole center for 7-hole disks and within the $\sim 1 \mu m$ -radius of the hole center for 3-hole disks. This requirement is fulfilled by the appropriate sample preparation.

Oxygen three isotope analyses of the magnesian CC chondrules from the SaU 290 CH chondrite showed that most of the chondrules have nearly identical oxygen isotope ratios ($\Delta^{17}O = -2.2 \pm 0.6$ %; 2SD), which is consistent with those of magnesian CC chondrules in other metal-rich carbonaceous chondrites (Krot et al. 2010). These data support the view that most magnesian CC chondrules in metal-rich carbonaceous chondrites formed by a single impact heating event (Krot et al. 2010). However, the rest of the magnesian CC chondrules from SaU 290 have a wide range of $\Delta^{17}O$ values from -6.4 % to +2.2 % that obviously could not be formed by the impact heating event common to other magnesian CC chondrules with $\Delta^{17}O$ value of -2.2 %. Given that the range of $\Delta^{17}O$ values (-6.4 % to +2.2 %) is within (similar to) that of the type I porphyritic chondrules in a CH/CB chondrite (Krot et al. 2010), it may be possible that the CC chondrules with distinct $\Delta^{17}O$ values formed together with the type I porphyritic chondrules in the solar nebula.

ACKNOWLEDGEMENTS

The authors are grateful to K. Nagashima and P. R. Heck for their thorough and helpful reviews and A. Ruzicka for thoughtful comments and for handling this paper. The authors also thank M. Kimura for constructive discussion, T. Nakamura and T. Noguchi for their helpful advice for sample preparation and support for SR-XRD, J. Fournelle for calibration and help with SEM and EPMA measurements, J. Kern for SIMS and profilometer support, and the Physical Sciences Laboratory at the University of Wisconsin for fabrication of multiple-hole disks. The bulk sample of SaU 290 was kindly provided from the Bartoschewitz Meteorite Laboratory. This work is supported by the NASA SRLIDAP program (NK, NNX09AC30G) and JSPS Research Fellowships for Young Scientists (DN). WiscSIMS is partly supported by NSF-EAR (0319230, 0744079).

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Figure captions

- **Fig. 1:** A photograph of the 7-hole disk (disk-1) mounted in the CAMECA holder (a), a photograph of the 3-hole disk (disk-3), schematic illustrations showing mounting an epoxy disk in the 7-hole disk (c) and epoxy disk preparation procedures (d). In panel (c), an upper disk without a tungsten lip is depicted (WiscSIMS design), while that used in Nakamura et al. (2008a) has a tungsten lip.
- Fig. 2: A photomicrograph of the >300 chondrules extracted from the SaU 290 CH chondrite.
- **Fig. 3:** Values of δ^{18} O at outer holes relative to the bracketing analyses at the center hole in the 7-hole disks (a, the disk-1; b, the disk-2). The error bars are internal error from counting statistics (2SE).
- **Fig. 4:** Values of δ^{18} O at two south holes relative to the bracketing analyses at the north hole in the 3-hole disks (a, the disk-3; b, the disk-4). The error bars are internal error from counting statistics (2SE).
- Fig. 5: Contour maps of all combined δ¹⁸O values of different locations relative to the bracketing analyses at the center of the grains in individual holes in the 7-hole disks (a) and in the 3-hole disks (b). Contour interval is 0.25 ‰. Small open circles indicate locations where oxygen two isotopes were analyzed (56 spots in the 7-hole disks and 80 spots in the 3-hole disks).
- **Fig. 6:** BSE images of polished surfaces of the SaU 290 chondrules. In panel b, a brighter grain next to the chondrule Ch02 is San Carlos olivine. In panel d, three holes in the chondrule Ch05 are pits made by oxygen isotope analyses with 15 μm spots. In panel f, white dots scattering on the chondrule surface are Fe-metal grains (< 1 μm in diameter; detected by SEM-EDS). In panel h, there appears to be a dark rim surrounding the porous interior. This is just due to an inclined periphery and is not a feature of the chondrule. Oval regions with white circles outline pits from oxygen isotope measurements (Ch09 was not analyzed with 15 μm spots).
- **Fig. 7:** Enlarged BSE images of the SaU 290 chondrules. In panel b, white dots scattering on the chondrule surface are Fe-metal grains (< 1 μm in diameter; identified by SEM-EDS). White-outlined ovals represent the primary beam size of 3 × 4 μm for reference, but do not indicate sputtered areas. The ellipses in panels a and d indicate mineral phases but not locations of oxygen isotope analyses.

- Abbreviations: Hpx, high-Ca pyroxene; Lpx, low-Ca pyroxene; glass, feldspathic glass; Wo5, one of two types of low-Ca pyroxene (En94Wo5) in Ch06; Wo1, one of two types of low-Ca pyroxene (En98Wo1) in Ch06; Cr, cristobalite.
- **Fig. 8:** Synchrotron radiation X-ray diffraction patterns of the SaU 290 chondrules in a range of diffraction angles (2θ) from 4 to 62 °.
- **Fig. 9:** Defocused-beam (~ 25 μm) electron microprobe analyses of the SaU 290 chondrules in terms of atomic ratios of Al, (Mg+Fe), and Si. Also shown are stoichiometric olivine, low-Ca pyroxene, feldspar (anorthite, albite, and orthoclase), and silica.
- **Fig. 10:** Bulk chemical compositions (by defocused-beam EPMA) of the SaU 290 chondrules normalized by Si and elemental abundance of CI chondrites (Anders and Grevesse 1989) in linear scale. The dotted area represents a field of bulk chemical compositions of CC chondrules in other CH chondrites (Krot et al. 2000).
- **Fig. 11:** Relative δ^{18} O and δ^{17} O values of outer SC-Ol grains from the analyses at the center SC-Ol grain (disk-1). In individual holes, the SC-Ol grains were used as running standards bracketing chondrule analyses. Error bars are external reproducibility (2SD). TF represents the terrestrial fractionation line.
- **Fig. 12:** Oxygen isotope ratios of the SaU 290 chondrules analyzed with 15 μm spots (a) and with 4 μm spots (b). TF, Y&R, and CCAM represent the terrestrial fractionation line, the Young & Russell line (Young and Russell 1998), and the carbonaceous chondrite anhydrous mineral line (Clayton et al. 1977). Oxygen isotope ratios of the bulk meteorite are from Park et al. (2005).
- **Fig. 13:** Oxygen isotope ratios of the SaU 290 chondrules are plotted as deviations from the terrestrial fractionation (TF) line ($\Delta^{17}O = \delta^{17}O 0.52 \times \delta^{18}O$).

Table 1. Measured δ^{18} O values relative to SC-Ol grain in the center hole*

disk-1	$\delta^{18}O \pm 2$	2SE (‰)	disk-2	$\delta^{18}O \pm 2$	2SE (‰)
	0.51	0.37		0.08	0.32
NW	0.42	0.38		-0.14	0.33
IN W	0.32	0.38	W	0.05	0.32
	0.42	0.19		-0.15	0.27
				-0.04	0.25
	-0.31	0.41		-0.39	0.27
W	-0.51	0.38		-0.38	0.31
**	-0.49	0.39	NW	-0.51	0.31
	-0.44	0.22		-0.54	0.31
				-0.46	0.16
	0.71	0.39		-0.34	0.28
NE	NE 0.53 0.41 -0.33 0.41	0.26			
IVL	-0.07	0.41	NE	-0.38	0.31
	0.39	0.82		-0.47	0.32
				-0.38	0.13
	0.85	0.32		-0.19	0.27
Е	0.46	0.32		-0.10	0.30
L	0.55	0.39	E	-0.20	0.29
	0.62	0.41		-0.22	0.33
				-0.18	0.11
	-0.41	0.37	_	0.26	0.31
SW	-0.63	0.38		0.17	0.27
~	-0.82	0.39	SE	0.13	0.27
	-0.62	0.41	_	-0.03	0.25
				0.13	0.25
	0.04	0.22		0.05	0.20
	0.04	0.33	4	-0.35	0.29
SE	-0.28	0.39	CIV	-0.42	0.30
	-0.22	0.41	SW	-0.59	0.33
	-0.16	0.34		-0.38	0.32
				-0.43	0.22
~			-		
Center	1 1	0.3	Center age values as	1 .	0.22

^{*}Bold-faced numbers are average values and external reproducibility (2SD).

Table 2. Measured d¹⁸O values relative to SC-Ol grain in the north hole*

disk-3	$\delta^{18}O \pm 2SE$ (‰)		disk-4	$\delta^{18}O \pm 2SE (\%)$	
	-0.08	0.22		-0.61	0.32
	0.00	0.23		-0.62	0.29
SW	0.06	0.21	SW	-0.88	0.29
	-0.03	0.24		-0.70	0.27
	-0.02	0.12		-0.70	0.24
	-0.31	0.22		0.46	0.31
	-0.42	0.23		0.07	0.27
SE	-0.39	0.22	SE	0.01	0.28
	-0.65	0.23		0.04	0.34
	-0.44	0.29		0.15	0.43

^{*}Bold-faced numbers are average values and external reproducibility (2SD).

Table 3. Averaged major elemental compositions of the SaU 290 chondrules obtained by defocused-beam electron microprobe analyses and transparency of the chondrules

chd#	Ch01	Ch02	Ch04	Ch05	Ch06	Ch07	Ch08	Ch09*
Number of Analyses	n=22	n=25	n=8	n=5	n=9	n=8	n=4	n=1
SiO ₂	55.3	55.1	54.0	56.0	62.7	53.7	55.4	55.0
TiO2	0.27	0.08	0.08	bdl	0.16	0.07	bdl	0.32
Al ₂ O ₃	6.08	1.35	1.07	0.17	3.82	1.25	0.55	3.30
Cr2O3	0.57	0.70	0.70	0.60	0.71	0.72	0.80	0.60
FeO	0.69	0.32	0.46	0.14	0.40	1.09	0.24	0.45
MnO	0.24	0.11	0.11	bdl	0.26	bdl	0.12	0.21
MgO	31.6	41.2	41.9	42.2	27.2	41.6	41.4	35.0
CaO	3.91	1.24	0.91	0.12	2.59	1.14	0.45	2.38
NiO	bdl	bdl	bdl	bdl	bdl	bdl	bdl	bdl
Na ₂ O	1.07	bdl	bdl	bdl	0.60	bdl	bdl	bdl
K ₂ O	0.39	bdl	bdl	bdl	0.20	bdl	bdl	bdl
SO ₃	0.07	bdl	bdl	bdl	bdl	bdl	bdl	bdl
Total	100.2	100.1	99.3	99.3	98.7	99.7	99.0	97.4
Mg#	98.7	99.8	99.6	99.8	99.0	98.9	99.7	
Transparency [#]	Op	Op	Wh/Tr	Wh/Tr	Wh/Tr	Op	Wh/Tr	Wh/Tr

^{*}Broad beam EPMA data for Ch09 were not available, because of high porosity of Ch09. Instead, focused beam EPMA data with highest analytical total are listed, because analytical totals of other EPMA data are low due to high porosity (81-97 wt%).

bdl = below detection limit.

^{*}Transparency under an optical microscope. Abbreviations: Op, black opaque; Wh/Tr, white-clouded/transparent.

Table 4. Averaged major elemental compositions of individual mineral phases in the SaU 290 chondrules obtained by focused-beam electron microprobe analyses.

chd#	Ch01		Ch	02	Ch06			
	I C	High-Ca	Cl	Douls	Dericales	Low-Ca px		C:1:
	Low-Ca px	px	Glass	Dark	Bright	Wo1*	Wo5 [*]	Silica
Number of analyses	n=3	n=3	n=3	n=3	n=3	n=3	n=5	n=4
SiO ₂	50.73	53.28	54.20	52.76	53.75	59.55	60.23	79.41
TiO ₂	0.19	0.45	bdl	0.07	0.10	0.16	0.22	0.07
Al ₂ O ₃	3.49	5.45	5.99	1.25	1.34	1.26	2.51	2.39
Cr2O3	0.61	0.80	bdl	0.39	0.71	0.64	0.75	0.22
FeO	0.87	0.72	0.67	0.21	0.27	0.48	0.66	0.18
MnO	0.31	0.29	0.27	bdl	0.06	0.19	0.32	0.08
MgO	42.64	27.12	32.93	42.88	41.46	36.35	32.78	8.86
CaO	1.88	11.04	3.06	0.66	0.94	0.59	2.59	1.21
Na ₂ O	0.34	0.27	1.04	bdl	bdl	bdl	0.07	0.41
K ₂ O	bdl	bdl	0.43	bdl	bdl	bdl	bdl	0.11
Total	101.06	99.43	98.59	98.30	98.61	99.23	100.12	92.95
bdl = below detection limit.								
*See Fig. 7.								

Table 5. Oxygen isotope ratios of the SaU 290 chondrules measured with 15 μm spots, using the 7-hole disk (disk-1) and IMS-1280 SIMS*

	$\delta^{18}O_{VSMOW}$ ±	2SD (‰)	$\delta^{17}O_{VSMOW}$ ±	2SD (‰)	$\Delta^{17}O \pm 2S$	SD (‰)
	7.03	0.37	5.62	0.86	1.96	0.77
Ch01	7.40	0.37	5.94	0.86	2.09	0.77
	7.05	0.37	6.09	0.86	2.42	0.77
	0.97	0.74	-1.81	0.53	-2.31	0.55
Ch02	0.75	0.74	-1.98	0.53	-2.36	0.55
	0.75	0.74	-1.68	0.53	-2.07	0.55
	0.01	0.20	-2.20	0.56	-2.21	0.52
Ch04	-0.33	0.20	-2.27	0.56	-2.10	0.52
	0.41	0.20	-2.55	0.56	-2.77	0.52
	0.51	0.81	-1.81	0.69	-2.07	0.82
Ch05	0.76	0.81	-1.64	0.69	-2.03	0.82
	0.57	0.81	-1.55	0.69	-1.84	0.82
	-6.19	0.46	-9.07	0.65	-5.85	0.55
$\mathrm{Ch06}^{^{\#}}$	-6.64	0.46	-10.18	0.65	-6.73	0.55
	-6.14	0.46	-9.95	0.65	-6.76	0.55
	0.64	1.85	-2.17	1.60	-2.50	0.85
Ch07	0.87	1.85	-2.28	1.60	-2.73	0.85
	0.81	1.85	-1.37	1.60	-1.79	0.85
	0.20	0.71	-2.36	0.88	-2.46	0.73
Ch08	0.14	0.71	-1.75	0.88	-1.83	0.73
	0.11	0.71	-2.54	0.88	-2.60	0.73

^{*}The uncertainties associated with isotope ratios are external reproducibility (2SD) of six to eight sets of bracketing analyses of San Carlos olivine standard grains.

[#]For Ch06 containing silica, instrumental bias was corrected as a mixture of low Ca pyroxene and silica (14 %), which resulted in δ^{18} O shift of -1.1 % in the bias correction factor compared to low Ca-pyroxene.

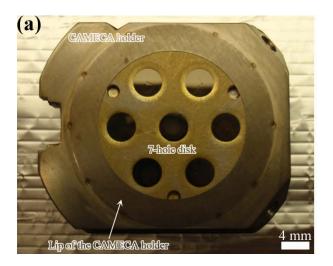
Table 6. Oxygen isotope ratios of pyroxene grains in the SaU 290 chondrules measured with 4 μ m spots, using the 7-hole disk (disk-1) and IMS 1280-SIMS^{*,#}

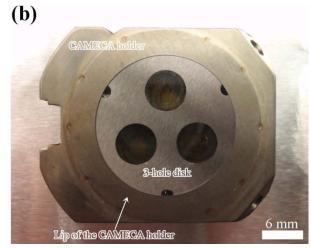
	$\delta^{18}O_{VSMOW}$	± 2SD (‰)	$\delta^{17}O_{VSMOW} =$	= 2SD (‰)	$\Delta^{17}O \pm 23$	SD (‰)	Remarks ^{\$}
	8.98	0.70	6.60	0.98	1.93	0.87	low-Ca px
	8.11	0.70	5.60	0.98	1.38	0.87	low-Ca px
Ch01	8.25	0.70	5.60	0.98	1.31	0.87	high-Ca px
	7.78	0.70	5.80	0.98	1.75	0.87	high-Ca px
	7.96	0.70	6.72	0.98	2.58	0.87	high-Ca px
	0.83	0.66	-1.22	0.72	-1.65	0.89	brighter phase
	1.58	0.66	-2.60	0.72	-3.42	0.89	brighter phase
Ch02	1.26	0.66	-1.31	0.72	-1.96	0.89	darker phase
	1.12	0.66	-2.33	0.72	-2.91	0.89	darker phase
	1.51	0.66	-1.17	0.72	-1.96	0.89	darker phase
	1.36	0.87	-1.36	0.83	-2.07	0.74	
	1.22	0.87	-0.94	0.83	-1.57	0.74	
Ch04	0.36	0.87	-1.05	0.83	-1.24	0.74	
	0.88	0.87	-1.55	0.83	-2.00	0.74	
	1.90	0.87	-1.10	0.83	-2.08	0.74	
	-5.75	1.77	-9.27	1.58	-6.28	1.05	Wo5
	-6.64	1.77	-9.57	1.58	-6.12	1.05	Wo5
	-6.03	1.77	-9.82	1.58	-6.68	1.05	Wo1
C1.06	-8.51	1.77	-11.15	1.58	-6.73	1.05	Wo1
Ch06	-8.57	1.77	-10.04	1.58	-5.58	1.05	Wo1
	-10.90	1.77	-12.53	1.58	-6.86	1.05	Wo5
	-7.66	1.77	-10.81	1.58	-6.83	1.05	Wo5
	-11.05	1.77	-12.44	1.58	-6.69	1.05	Wo1

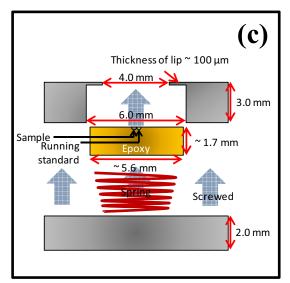
 $^{^*}$ Ch09 was also analyzed with 4 μ m spots, but the 16 O count rates were only 60 % of the SC-Ol grain and other chondrules. Therefore, we decided the Ch09 data were unreliable and do not list in the table. Instead, Ch09 data is listed in online supplements.

[#]The uncertainties associated with isotope ratios are external reproducibility (2SD) of six sets of bracketing analyses of San Carlos olivine standard grains.

^{\$}See Fig. 7.







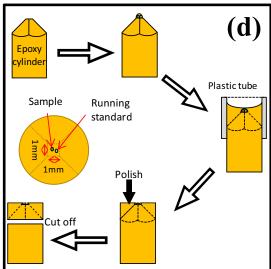


Fig. 1: Nakashima et al. 2010

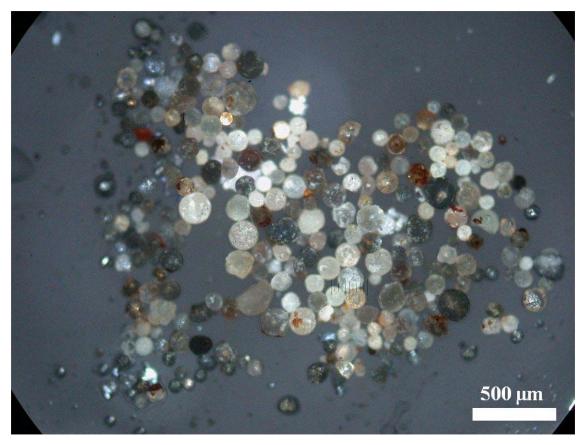


Fig. 2: Nakashima et al. 2010

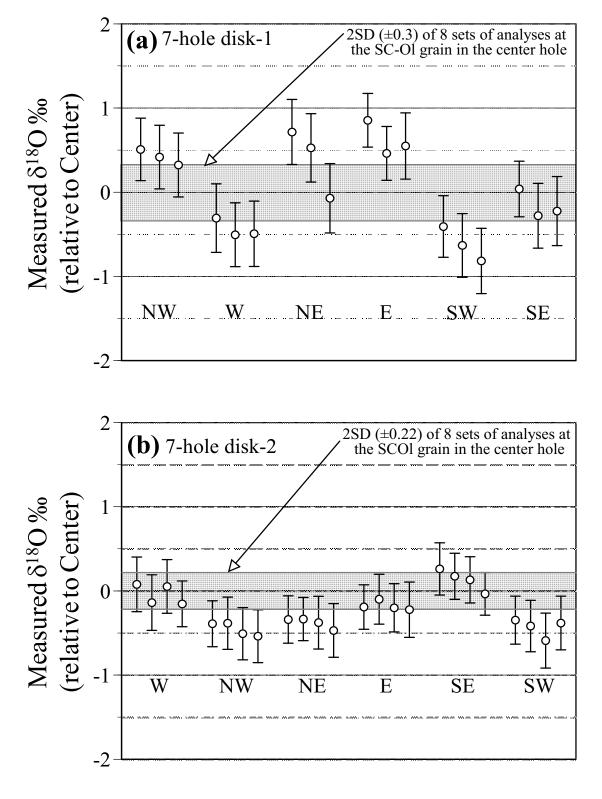


Fig. 3: Nakashima et al. 2010

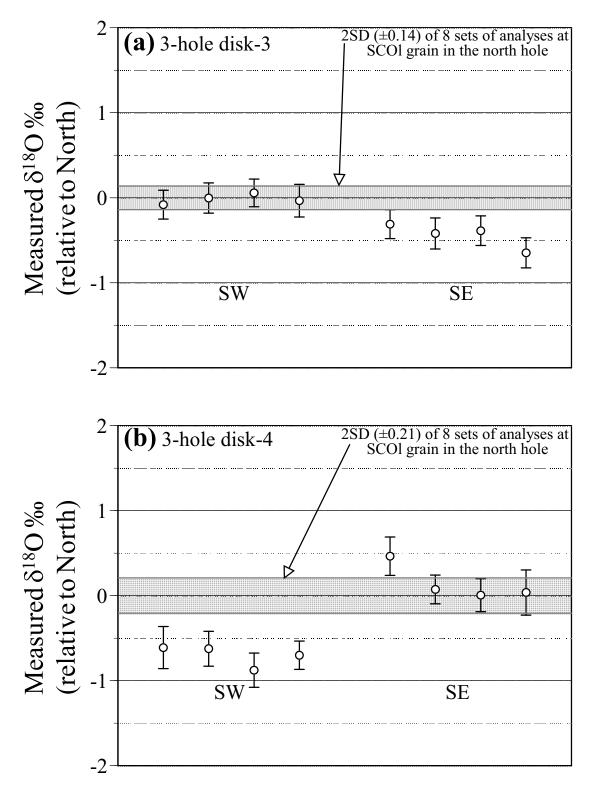
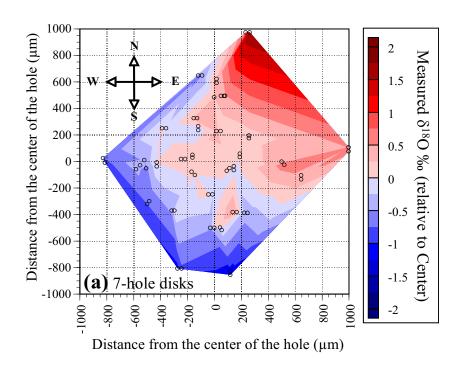


Fig. 4: Nakashima et al. 2010



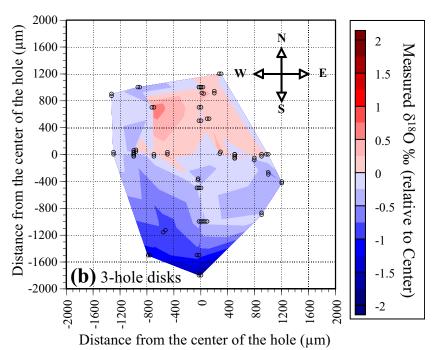


Fig. 5: Nakashima et al. 2010

11

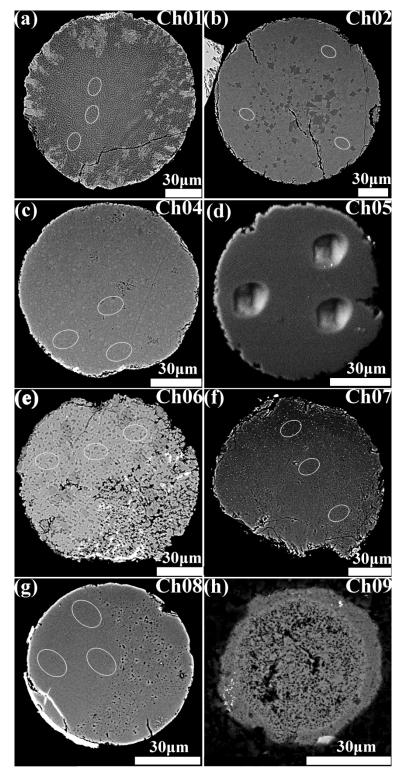


Fig. 6: Nakashima et al. 2010

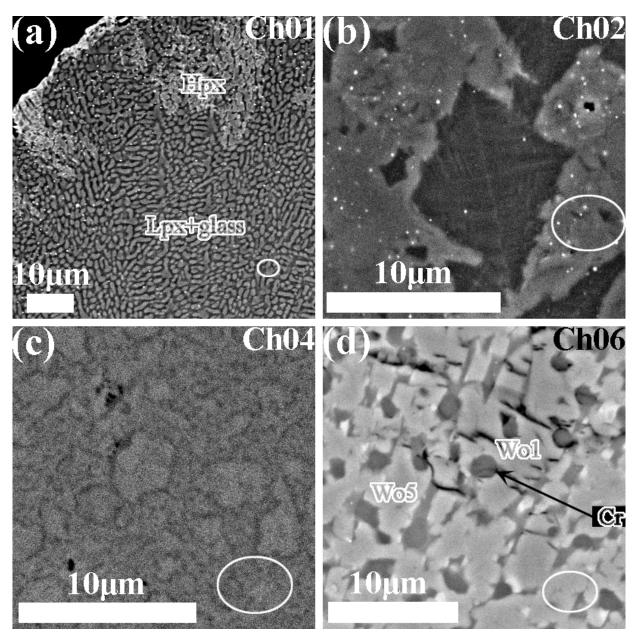


Fig. 7: Nakashima et al. 2010

Ol: Olivine

Px: low-Ca pyroxene

Cr: Cristobalite

Un: a phase unidentified

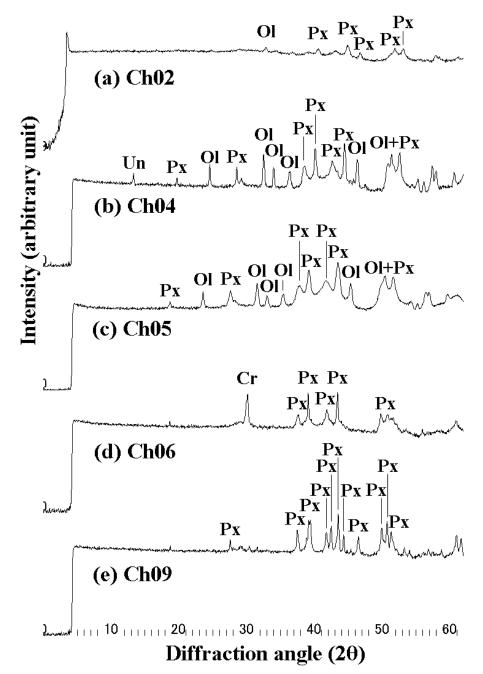


Fig. 8: Nakashima et al. 2010

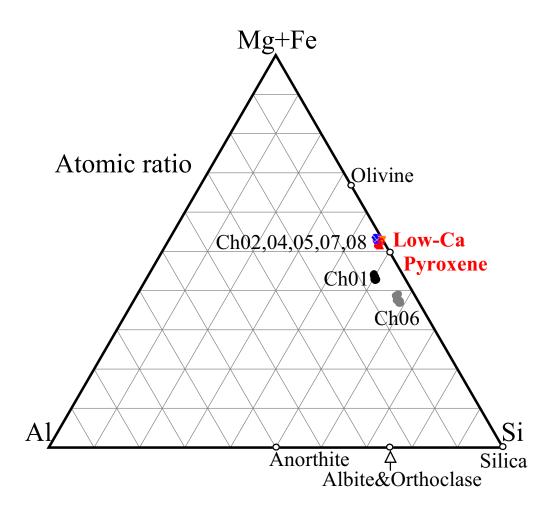


Fig. 9: Nakashima et al. 2010

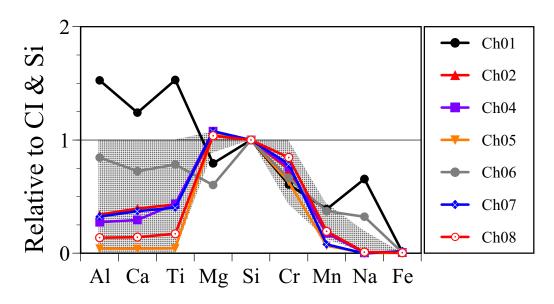


Fig. 10: Nakashima et al. 2010

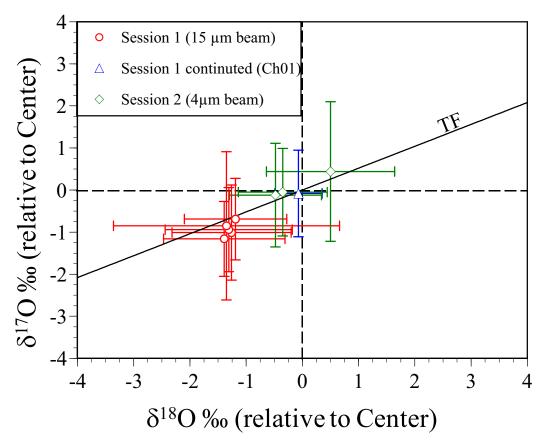


Fig. 11: Nakashima et al. 2010

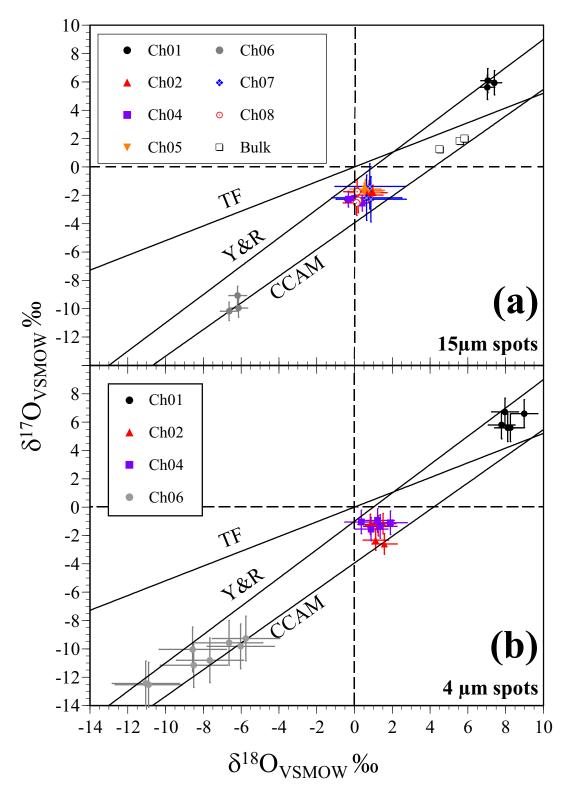


Fig. 12: Nakashima et al. 2010

18

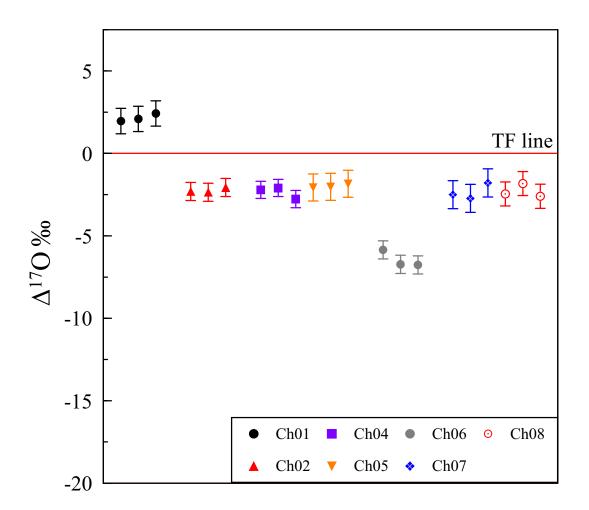


Fig. 13: Nakashima et al. 2010

Table A1. Measured $\delta^{18}O$ values relative to the center of the grains in individual holes in the 7-hole disk-1*

Disk-1		s180 + 20	IF (0/) -	ΔX	ΔY
		$\delta^{18}O \pm 2S$	E (‰) =	(µn	1)
		-0.38	0.39	247	-387
	S	-0.30	0.37	223	-387
		-0.34	0.11		
		0.03	0.41	190	59
	E	0.04	0.37	190	34
Center		0.04	0.02		
(± 0.32 ; 2SD)		0.07	0.38	-169	-77
	W	0.00	0.38	-144	-102
		0.04	0.09		
	N	0.27	0.40	-153	327
		0.51	0.38	-128	327
		0.39	0.34		
	~	-0.11	0.30	-296	-370
	S	-0.45	0.33	-316	-370
		-0.28	0.48		
		0.15	0.32	-162	50
	W	0.21	0.32	-162	30
West		0.18	0.08		
(± 0.21 ; 2SD)					
	N	0.15	0.28	258	196
	17	0.13	0.34	258	176
		0.14	0.02		
		0.16	0.29	113	-50
	E	0.11	0.29	93	-70
		0.14	0.07		

^{*}Bold-faced numbers are average values and external reproducibility (2SD).

Table A2. Measured $\delta^{18}O$ values relative to the center of the grains in individual holes in the 7-hole disk-2*

Disk-2		$\delta^{18}O \pm 2S$	E (%) -	ΔΧ ΔΥ	
2101	-			(µm)	
		0.02	0.32	-427	-7
	W 500μm	-0.06	0.31	-427	-37
		-0.02	0.11		
		-0.08	0.31	146	-35
	E 150µm	-0.06	0.33	146	-65
	•	-0.07	0.02		
Center (± 0.23 ; 2SD)		-0.23	0.33	-13	-248
(± 0.23 , 23D)	S 250µm	0.04	0.33	-43	-248
	5 250µm	-0.09	0.28	-43	-240
				• • • •	
		0.05	0.33	-388	251
	N 500μm	-0.20	0.32	-358	251
		-0.07	0.35		
		0.21	0.47	-247	19
	W 500μm	-0.01	0.47	-217	19
	W 300μm	0.10	0.31	-21/	1,
		0.10	0.31		
	E 1000μm	0.58	0.45	998	77
		0.51	0.49	998	107
		0.54	0.10		
Northwest	N 500	-0.60	0.46	-120	649
$(\pm 0.42; 2SD)$		-0.60 -0.44			
	N 500μm		0.51	-90	649
		-0.52	0.23		
		0.46	0.47	137	-381
	S 500µm	0.21	0.46	167	-381
	•	0.34	0.35		
		0.44	0.40	701	
	W 500um	-0.44	0.49	-581	-58
	W 500μm	-0.33	0.48	-551	-28
		-0.39	0.16		
		0.46	0.50	646	-104
	E 500µm	0.43	0.47	646	-134
		0.44	0.04		
Southwest (+0.44: 2SD)		0.41	0.49	17	593
(± 0.44 ; 2SD)	N 500µm	0.41	0.49	17	623
	14 200μπ	0.27	0.31	1 /	023
			· · -		
		-1.26	0.49	119	-855
	S 500µm	-1.09	0.50	119	-825
		-1.17	0.23		
			Cont	inued on ne	xt page

21

Table A2. Cor	ntinued				
		0.39	0.34	50	495
	N 500μm	0.49	0.37	80	495
	·	0.44	0.13		
		1.71	0.39	263	974
	N 1000μm	0.72	0.36	233	974
_	·	1.22	1.40		
East (± 0.28 ; 2SD)		0.05	0.34	0	-500
, , ,	S 500µm	-0.48	0.36	-30	-500
	•	-0.21	0.75		
		-0.79	0.38	-243	-807
	S 800µm	-1.06	0.34	-273	-807
		-0.92	0.38		
		-0.07	0.27	43	-498
	S 500µm	-0.10	0.30	55	-517
	•	-0.09	0.05		
		0.42	0.27	500	0
	E 500µm	0.62	0.34	521	-24
		0.52	0.28		
		0.25	0.32	-1	485
West	N 500µm	0.46	0.31	73	495
(± 0.22 ; 2SD)		0.35	0.29		
		-0.47	0.31	-520	11
	W 500μm	-0.56	0.33	-506	-51
		-0.52	0.13		
		-0.74	0.38	-826	26
	W 1000μm	-0.84	0.41	-813	-8
		-0.79	0.15		
		-0.72	0.36	-483	-299
	S 300µm	-0.75	0.34	-499	-321
		-0.73	0.04	.,,	021
Southeast (± 0.30 ; 2SD)		0.30	0.36	17	230
(, ,	N 300µm	0.06	0.36	47	230
		0.18	0.35	•	
		0.00	0.28	110	267
Northeast (±	NW 200μm	0.00 -0.16	0.28	-119	267
0.19 ; 2SD)	19 W 200μ111	-0.16 - 0.08	0.27 0.23	-119	237
*Bold-faced n	umbers are ave			nal reprodu	cibility

^{*}Bold-faced numbers are average values and external reproducibility (2SD).

Table A3. Measured $\delta^{18}O$ values relative to the center of the grains in individual holes in the 3-hole disk-3*

Disk-3		$\delta^{18}O \pm 2SE (\%)$ -		ΔX	ΔΥ
Disk	<u> </u>	0 O ± 28	DE (700) -	(µm)	
		-0.05	0.35	0	500
	N 500μm	0.06	0.34	-30	500
		0.00	0.16		
		0.31	0.32	0	1000
	N 1000μm	0.20	0.34	30	1000
		0.25	0.15		
		-0.25	0.35	-30	-500
	S 500µm	-0.29	0.35	-60	-500
		-0.27	0.05		
		-0.60	0.33	0	-1000
	S 1000µm	-0.41	0.37	-30	-1000
		-0.50	0.26		
		-0.89	0.36	-30	-1500
North	S 1500µm	-0.86	0.34	-60	-1500
(±0.29; 2SD)		-0.88	0.05		
		-1.81	0.34	0	-1800
	S 1800µm	-1.63	0.33	-30	-1800
		-1.72	0.25		
		0.34	0.33	-500	0
	W 500μm	0.12	0.35	-500	30
		0.23	0.30		
		0.33	0.35	500	0
	E 500µm	0.46	0.35	500	-30
		0.40	0.19		
		0.18	0.40	900	0
	E 900µm	-0.09	0.34	900	-30
		0.04	0.37		

Continued on next page

Table A3. Co	ntinued					
		0.16	0.35	90	530	
	N 500µm	0.21	0.36	120	530	
		0.18	0.07			
		0.04	0.38	197	940	
	N 1000μm	0.16	0.37	197	910	
	·	0.10	0.17			
		-0.18	0.36	499	1	
	E 500µm	-0.08	0.37	499	-59	
	·	-0.13	0.15			
		-0.09	0.33	792	-53	
	E 800µm	-0.28	0.37	792	-83	
	·	-0.19	0.28			
		-0.19	0.38	0	-500	
Southwest	S 500µm	-0.12	0.38	-30	-500	
$(\pm 0.31; 2SD)$	·	-0.15	0.10			
		-0.28	0.37	60	-1000	
	S 1000µm	-0.19	0.39	90	-1000	
	·	-0.24	0.12			
		-0.51	0.41	-1000	30	
	W 1000μm	-0.35	0.34	-1000	60	
		-0.43	0.23			
		-0.38	0.35	-940	1000	
	NW 1000μm	-0.51	0.37	-910	1000	
		-0.45	0.19			
		-0.02	0.37	900	-900	
	SE 900μm	-0.40	0.35	900	-870	
		-0.21	0.54			
		Continued on next page				

Table A3. Con	ntinued				
		-0.21	0.31	500	-30
	E 500µm	-0.10	0.34	500	-60
		-0.15	0.16		
		-0.57	0.29	1000	-270
	E 1000µm	-0.31	0.31	1000	-300
	2 1000 1	-0.44	0.36	1000	200
		-0.47	0.33	1200	-430
	E 1200µm	-0.47	0.33	1200	-405
	2 1200 pm	-0.47	0.01	1200	105
		-0.01	0.28	-700	0
	W 700μm	-0.16	0.31	-700	-30
	11 700 min	-0.09	0.21	700	30
		-0.07	0.32	-1000	0
	W 1000μm				0
	w 1000µm	-0.19	0.35	-1000	-30
		-0.13	0.17		
		-0.08	0.33	-1300	0
Southeast	W 1300μm	-0.03	0.31	-1300	30
$(\pm 0.25; 2SD)$		-0.06	0.08		
		0.05	0.32	-1330	900
	NW 1300μm	-0.13	0.33	-1330	870
		-0.04	0.25		
		0.08	0.35	0	1000
	N 1000µm	-0.25	0.32	-30	1000
		-0.09	0.47		
		0.27	0.32	270	1200
	NE 1200μm	-0.12	0.35	300	1200
		0.08	0.55		
		-0.92	0.32	-530	-1130
	SW 1200μm	-0.81	0.30	-560	-1160
	•	-0.86	0.16		
		-1.12	0.34	-757	-1502
	SW 1600µm	-1.12	0.34	-737 -787	-1502
	5 11 1000μ111	-1.10 -1.11	0.30	-/0/	-1302
17.110.1		-1.11	0.02		

^{*}Bold-faced numbers are average values and external reproducibility (2SD).

Table A4. Measured $\delta^{18}O$ values relative to the center of the grains in individual holes in the 3-hole disk-4*

III IIIuIviuuai i	ioles in the 3-n	ole disk-4			
		-0.23	0.22	-1000	0
	W 1000μm	0.05	0.23	-1000	30
		-0.09	0.40		
		0.04	0.27	1000	0
	E 1000µm	-0.04	0.24	970	0
	2 1000 pm	0.00	0.11	770	Ü
North		0.00	0,11		
(±0.16; 2SD)		0.14	0.24	0	1000
(-0.10, 252)	N 1000µm	0.18	0.25	-30	1000
	1, 1000 mil	0.16	0.06	50	1000
		0.10	0.00		
		-0.48	0.24	0	-1000
	S 1000µm	-0.25	0.21	30	-1000
	В тобории	-0.36	0.32	30	1000
		-0.50	0.52		
		0.43	0.44	12	913
	N 1000μm	0.43	0.43	38	904
	14 1000μπ	0.43	0.00	30	704
		0.43	0.00		
		0.09	0.42	-967	66
	W 1000μm	-0.18	0.41	-966	41
	** 1000piii	-0.15	0.39	-700	71
Sothwest		-0.03	0.37		
(±0.37; 2SD)		0.33	0.40	290	36
$(\pm 0.57, 25D)$	E 300µm	-0.12	0.46	279	10
	2 300µm	0.12	0.63	217	10
		0.10	0.03		
		-0.41	0.40	-42	-363
	S 400µm	-0.22	0.41	-48	-384
	В 400µш	-0.22	0.41	-40	-304
		-0.32	0.20		
		-0.57	0.37	-1000	0
	W 1000μm	-0.66	0.34	-1000	-30
	νν 1000μπ	-0.62	0.13	-1000	-30
		-0.02	0.13		
Southeast		-0.04	0.36	0	700
(±0.26; 2SD)	N 700µm	0.09	0.35	-30	700
(±0.20, 23D)	11 /00μ111	0.09	0.33	-30	700
		0.04	0.10		
		0.64	0.36	-700	700
	SW 700μm	-0.21	0.30	-700 -730	700
	5 7 7 7 0 0 μ 111	0.21 0.22		-/30	/00
15.44.0.4		0.22	1.20		

^{*}Bold-faced numbers are average values and external reproducibility (2SD).

Table A5. Oxygen isotope ratios of the chondrule Ch09 from SaU 290 measured with 4 μm spots, using the 7-hole disks (disk-1) and IMS-1280 $SIMS^*$

	$\delta^{18}O \pm 29$	SE (‰)	$\delta^{17}O \pm 2S$	SE (‰)	Δ^{17} O ± 2	SE (‰)	¹⁶ O (10 ⁶ cps)
	2.36	0.50	3.03	1.22	1.80	1.25	18.9
SC-O1	1.24	0.67	2.58	1.14	1.94	1.19	18.4
	1.28	0.60	1.78	1.12	1.12	1.16	18.2
	-2.19	0.93	-6.37	1.80	-5.24	1.86	9.0
Ch09	-3.56	0.78	-9.17	1.80	-7.32	1.84	10.7
CHOS	-8.86	0.88	-10.21	1.24	-5.60	1.32	11.1
	-11.47	0.77	-13.33	1.83	-7.37	1.88	11.0
	2.11	0.46	3.69	0.94	2.59	0.97	18.4
SC-Ol	1.99	0.52	2.81	1.11	1.78	1.14	17.9
SC-01	1.23	0.58	3.65	1.31	3.02	1.34	17.7
	1.70	1.02	2.92	1.43	2.04	1.34	18.3

^{*}Oxygen isotope ratios of Ch09 are corrected for instrumental bias. Bold-faced numbers are average values of bracketing standard analyses and external reproducibility (2SD).