Early to Late Holocene Contrast in productivity, OMZ Intensity and Calcite

Dissolution in the Eastern Arabian Sea

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Abstract

The Oxygen Minimum Zone (OMZ) is a potential source of CO₂ to the atmosphere and the

extensive OMZ spread to 1.37 x 10⁶ km² in the northern Arabian Sea would be an important

contributor to the CO2 budget. In this perspective, we report here that the eastern Arabian Sea

experienced coeval increase in productivity and denitrification from ~7 to 0 kyr, which coincides

with minimum dissolved oxygen levels in bottom-waters. In addition, during the same period, an

increased calcium carbonate dissolution occurred as evidenced by low CaCO₃ content, decreasing

shell weights of planktonic foraminifer species Globigerinoides ruber with well-marked dissolution

features on their shells. Therefore, this study has implications on the role of OMZ in governing

CaCO₃ dissolution and contributing to an increase in atmospheric CO₂.

Keywords: Productivity, denitrification, OMZ, calcite dissolution, Arabian Sea

Introduction

The Arabian Sea is unique in terms of seasonal reversals of winds during the south-west (SW) and north-east (NE) monsoon due to seasonal variations in heating of the southern Asian continent. Associated with the seasonal reversal of winds is the reversal of surface-water circulation. Strong winds blowing from the SW generate upwelling of nutrient-rich waters in the Western Arabian Sea (WAS), off Somalia/Oman leading to very high primary production, and weak and sporadic upwelling along the coasts off India and Pakistan (Eastern Arabian Sea (EAS)) (Fig. 1). The EAS receives more precipitation compared to the WAS during the SW monsoon. Most of the SW monsoon precipitation on the Western Ghats drains into the Arabian Sea through numerous rivers and streams. During the NE monsoon winds from the NE evaporate and cool the surface waters which deepens the mixed-layer and leads to input of nutrients to the surface in the Northern basin and also increases the primary productivity (Madhupratap et al., 1996). The organic production in the Arabian Sea is however significantly higher during the SW monsoon in comparison to the NE monsoon and decreases in the intermonsoonal seasons.

The Arabian Sea is amongst the highest seasonal productivity zones in the world (Qasim, 1977). The high biological productivity prevailing during the south-west monsoon is a source of large amounts of organic matter which subsequently sinks to deeper depths (Nair et al., 1989). At intermediate depths, the oxidation of organic matter consumes dissolved oxygen and therefore the entire Arabian Sea is characterized by a permanent, intense Oxygen Minimum Zone (OMZ) between depths of 150-1200m (Wyrtki, 1971; Von Stackelberg, 1972). In the OMZ, bacterial degradation of organic matter takes place, which reduces NO₃⁻ and produces N₂ and N₂O in the water-column by a chain of processes known as denitrification (Deuser et al., 1978; Naqvi, 1991). OMZs are important due to their climatic interactions and impacts on ecosystems for eg., their contribution to major greenhouse gases such as CO₂ and N₂O (Resplandy et al., 2012). Modern observations suggest that the volume of OMZs' suboxic waters has increased over the past decades (Stramma et al., 2008) and could further expand due to ocean warming and climate change (Sarmiento et al., 1998; Keeling et al., 2010). Therefore, improving our understanding of OMZs is of primary importance when considering the future evolution and impacts of OMZs (Resplandy et al., 2012). Several studies were carried out to understand variations in productivity/denitrification and OMZ intensity in the EAS focussed on glacial-interglacial or longer timescales (Schulz et al., 1998; Reichart et al., 1998; Schulte et al., 1999; von Rad et al., 1999b; Ganeshram et al., 2000; Agnihotri et al., 2002; Reichart et al., 2004; Pichevin et al., 2007; Böning and Bard, 2009) and few studies covering shorter

timescales in the EAS (Von Rad *et al.*, 1999a; Agnihotri *et al.*, 2003; Agnihotri *et al.*, 2008). The OMZ is known to be intense during the Holocene, in particular during the Late Holocene (Gupta *et al.*, 2011) controlled by the supply of oxygen depleted intermediate waters combined with elevated monsoon-controlled oxygen demand (Böning and Bard, 2009).

Our understanding on the dissolution/preservation of calcite in response to varying intensity of the OMZ is very little. In the present study we have examined a sediment core from the EAS with high sedimentation, covering the Holocene period, in order to address the changes in productivity, denitrification, oxygenation of bottom waters and to understand the influence of OMZ on the dissolution/preservation of calcite.

Materials and Methods

Core AAS9/19 (14° 30.115'N; 73° 08.515'E; water depth of 367 m; Fig. 1.) was recovered from the OMZ region, much above the present day lysocline (~3800m; Peterson and Prell, 1985) as well as the present day Aragonite Compensation Depth (ACD) (von Rad *et al.*, 1999b). The chronology of this core was based on four AMS 14 C dates and covers 12.3 kyrs (Table 1). AMS 14 C dating was performed on monospecific samples of the planktic foraminiferal species *G. ruber* using the Tandem Accelerator at Leibniz Laborfür Altersbestimmung und Isotopenforschung, Christian-Albrechts-Universität, Kiel, Germany. Measured 14 C ages were converted to sediment ages using the online CALIB radiocarbon calibration programme version 7.0 (Stuiver *et al.*, 2005). Before calibration, radiocarbon dates were corrected for a reservoir effect ($\Delta R=125 \pm 37$ years) based on observations for the Indian Ocean (Dutta *et al.*, 2001).

All the samples were oven dried at 50° C; a portion of the sample was weighed and disaggregated by soaking in distilled water and then wet sieved through a >63µm sieve. For shell weight measurements, the coarse material was further sieved in the size range of 300-355 µm. Approximately thirty *G. ruber* (white) shells were picked from this size fraction under a Stereo-zoom binocular microscope and weighed on a microbalance (1σ precision: $\pm 1.5\mu$ g, n=10).

Isotopic analysis on *G. ruber* specimens from a size range of 250-350 μ m were done using a Finnigan MAT 252 mass spectrometer equipped with an automatic carbonate preparation device at the Universität Bremen, Germany. The mean external error and reproducibility(1s) of carbonate standard is better than $\pm 0.07\%$ and $\pm 0.05\%$ for δ^{18} O and δ^{13} C, respectively. Oxygen and carbon isotopic values are expressed as per mil (%) deviation Vienna Pee Dee belemnite (VPDB) standard, calibrated by NBS 18, 19 and 20 standards.

The carbon and nitrogen concentrations and their isotope values were measured at the Marine Stable Isotope Lab (MASTIL) at the National Centre for Antarctic & Ocean Research, Goa, India using an Isoprime Stable Isotope Ratio Mass Spectrometer in continuous-flow mode coupled with an EA (Isoprime, Vario Isotope Cube). The external precisions on δ^{13} C & δ^{15} N are $\pm 0.04\%$ & $\pm 0.07\%$ (1 σ standard deviation) respectively obtained by repeatedly running Cellulose (IAEA-CH-3) & Ammonium sulphate(IAEA-N1) standards (n=21). The external precisions for C% & N% are $\pm 0.39\%$ & $\pm 0.46\%$ respectively (1 σ standard deviation) obtained by repeatedly running Sulfanilamide as the standard (n=21). δ^{13} C values are reported with respect to V-PDB and δ^{15} N values are reported with respect to air N₂. The reference standard used for normalizing to V-PDB and air N₂ scale are Cellulose (IAEA-CH-3) & Ammonium sulphate (IAEA-N1).

For elemental analysis, about 50 mg of sediment was weighed in a Teflon beaker. To this 10 ml of acid mixture of HF+HNO₃+HClO₄ (7:3:1 ratio) was added and the contents were concentrated to a paste by placing the beaker on a hot plate. To this paste 4 ml of 1:1 HNO₃ was added. After heating for 5 min the material was diluted with ultrapure (18.2 M Ω) water to a final volume of 100 ml. Similar digestion procedure was followed for standard reference materials (MAG-1, SGR-1) and blanks. These solutions were analyzed for Ti, Cr and Mo on an inductively coupled plasma-mass spectrometer (Thermo X Series 2) at NIO, Goa. The accuracy compared to the standard reference material and the precision of the data based on duplicate analysis was better than $\pm 6\%$. The calcium carbonate content was determined with a coulometer (UIC, Inc-CM5130 acidification module). Analytical grade calcium carbonate was used as the standard reference material. The accuracy and precision of measurements based on analysis of duplicate samples and standard reference material were within $\pm 2\%$.

Results

The CaCO₃ content varied from 38 to 72% with higher values encountered during early Holocene (from ~12 to 7kyr) (Fig. 2a). The Organic Carbon (OC) content varied from 0.8 to 5% with low OC values during the early Holocene and a gradual increase beginning from ~7kyr (Fig. 2b). The core top values are within the range of OC values obtained from the continental margin sediments (Paropkari *et al.*,1992). N content varied from 0.1 to 0.7% with variations similar to that of OC ($r^2 = 0.87$, n=36)(Fig. 2d). C/N values range from 3.7 to 14.5 units (Fig. 2e), which are generally within those reported for marine sedimentary organic matter 8 ± 2 (Reichart *et al.*, 1997; Bhushan *et al.*, 2001). Higher values during the early Holocene could be due to contribution from terrestrial organic matter. δ^{13} Corg values range from -21 to -22 ‰ (Fig. 2f). A crossplot of δ^{13} Corg

versus C/N ratios shows all values in the range of marine organic matter (Meyers, 1994) (Fig.3) indicating that deposition of organic matter at this site has been predominantly of marine origin throughout the Holocene. δ^{15} N values range from 3.4 to 7 ‰ with lower values during the early Holocene and gradually increasing towards the Late Holocene (Fig. 2c). The δ^{15} N values of the top sections of the present core are in the range observed for the EAS (Agnihotri*et al.*, 2003). The δ^{13} C of *G. ruber* varied from 0.86 to 1.59 ‰ (Fig. 2g). Molybdenum concentrations varied from 1.3 to 9.7 ppm and Chromium concentrations varied from 137 to 214 ppm. Higher concentrations of Mo and Cr were noted during ~7-0 kyr. Ti showed a decreasing trend through the Holocene period with concentrations varying from 0.3 to 0.8wt% (Fig. 2h) and are positively correlated to CaCO₃ ($r^2 = 0.36$, n=36) suggesting that terrigenous dilution is not the cause for CaCO₃ variations. The shell weights of *Globigerinoides ruber* range from 13 to 19 µg (Fig 2m). Higher shell weights were noticed during Younger Dryas (YD) and early Holocene and then a decrease in shell weights was noticed throughout the Holocene period. On an average the ~12-7 kyr shells weighed ~17 µg and the 7-0 kyr shells weighed ~14 µg, a ~3µg decrease in shell weights during the Late Holocene. Scanning Electron Micrographs (SEM) were taken on a 6360 LV, JEOL instrument.

Discussion

Changes in surface productivity, denitrification and bottom-water oxygenation intensity

Organic matter (OM) input is either generated by primary production in the photic zone or through transport of terrestrial OM to the marine realm. The organic carbon (OC) content of sediments depends upon various factors such as sedimentation rates, oxygen exposure times and organic matter composition (amount of refractory and non-refractory components) (see Zonneveld *et al.*, 2010). Studies on the OC distribution in surficial sediments collected along the western continental margin of India suggest that primary productivity is the major controlling factor (Calvert *et al.*, 1995). The AAS9/19 sediment core is located on the continental slope, characterised by the presence of fine material, high sedimentation rates, presence of OMZ and therefore high organic carbon content. The OC content shows higher values during the Late Holocene and appears to reflect surface water productivity variations.

The δ^{13} C of planktonic foraminifera can vary in response to productivity changes, carbonate chemistry of seawater and upwelling processes (Peeters *et al.*, 2002). In the EAS, the processes of upwelling and winter convective mixing aid in bringing ¹³C depleted subsurface waters to the surface, eventually depleting δ^{13} C of the surface waters. Planktonic foraminifera calcifying in these

waters record lower δ^{13} C in their shells. On the other hand an increase in surface productivity increases the δ^{13} C of surface seawater. Increasing productivity is also linked to upwelling of nutrient-rich subsurface waters, foraminifera taking in 12 C preferentially over 13 C and hence the organic matter gets enriched in 12 C. Therefore the δ^{13} C in shells is a net result of whichever process dominates. We observe lower values of δ^{13} C $_{G.ruber}$ and δ^{13} C $_{org}$ beginning from \sim 7 kyr and a positive relationship between the two parameters which also advocates the increase in productivity during this period. In summary, the Late Holocene period from \sim 7 to 0 kyr is dominated by higher productivity as seen from the high OC%, low δ^{13} C $_{org}$ and low δ^{13} C of planktic foraminifera as a result of increase in intensity of upwelling/winter convective mixing.

Productivity increase is linked to an increase in denitrification. Arabian Sea is one of the most important regions wherein water-column denitrification occurs. Denitrification results in enrichment of ¹⁵N in residual NO₃ and the lighter isotope (¹⁴N) is consumed relatively faster due to kinetics. The heavier NO₃⁻ is then supplied to the ocean surface via upwelling and gets incorporated in newly formed particulate organic matter. Thus, the nitrogen isotopic composition ($\delta^{15}N$) of particulate organic matter depositing in sediments has been used as a proxy for the intensity of water-column denitrification in the past (Altabet et al., 1995; 1999; Ganeshram et al., 1995). However, certain processes affect this proxy such as mixing with isotopically lighter terrestrial organic matter, organic matter alteration in water column and sediments, incomplete nitrate utilisation, variability in nitrogen fixation, etc. (See Ganeshram et al., 2000). We observed that organic matter deposited at this site is of marine origin and therefore temporal variations in $\delta^{15}N$ are not a result of mixing or dilution due to terrestrial organic matter. Furthermore, lack of correlation between $\delta^{13}C_{org}$ and $\delta^{15}N$ suggests that spatial variations in $\delta^{15}N$ are not due to mixing of terrestrial and marine organic matter (Peters et al., 1978). Significant positive relationships between $\delta^{15}N$ - OC and $\delta^{15}N$ - N ($r^2 = 0.86$, n=36) rule out any early diagenetic effect of organic matter degradation on the δ^{15} N signal (Agnihotri *et al.*, 2003). Incomplete nitrate utilisation, i.e. preferential uptake of 14N_{NO3}, by phytoplankton leading to increase in ¹⁵N in the remaining nitrate pool, and lowering of ¹⁵N in the sinking OM, is a process which is not of importance in the eastern Arabian Sea, due to near-complete utilization of available nutrients (Altabet et al., 1999). Enhancement of nitrogen fixation by cyanobacteria can produce isotopically lighter δ^{15} N as they mostly use atmospheric nitrogen with very little fractionation (See Ganeshram et al., 2000). However, in the eastern Arabian Sea, nitrogen fixation does not have much effect on ¹⁵N_{NO3-} (Altabet et al., 1999). δ¹⁵N values in AAS9/19 are significantly correlated with OC content ($r^2 = 0.8$, n=36) which suggests that subsurface denitrification is controlled by the extent of biological productivity. Thus, we infer that present core location has experienced intense denitrification during Late Holocene from \sim 7 to 0 kyr.

Variations in surface productivity are expected to induce changes in oxygen level of bottom waters/sediment-water interface due to organic matter degradation. Redox conditions of bottom waters/sedimentary column strongly influence the distribution of some of the trace elements, such as Mo and Cr amongst others. In anoxic basins, Cr (VI) appears to be rapidly reduced to the Cr (III) state and removed from solution. Mo lacks chemical reactivity under oxic conditions but is enriched to a great extent in anoxic conditions (see Calvert et al., 1993). In modern anoxic environments, Mo shows a strong positive correlation to OC (Algeo and Lyons, 2006) as seen in the present core ($r^2 =$ 0.48, n=36). Downcore variations of Mo and Cr, normalized to Al or Ti, are often used to infer paleo-redox conditions at the sediment-water interface at the time of deposition. Ti in marine sediment is mostly of continental origin and supplied through eolian and fluvial pathways. Cr/Ti and Mo/Ti both show higher values during the Late Holocene (Fig 2i and 2j) which suggests prevalence of sub-oxic waters during the Late Holocene. Continuous rise in denitrification and OMZ intensity through the Holocene was proposed to be due to the sluggish circulation of the Red Sea and Persian Gulf waters, which reached its present state at around 6 kyr and this prevented the oxygen-rich southern water masses to the northern Arabian Sea enabling a continuous build-up of oxygen deficiency (Pichevin et al., 2007).

Calcite dissolution due to variations in bottom-water oxygenation

Geochemical proxies such as OC, N, C/N, δ^{15} N, δ^{13} C_{org}, and δ^{13} C_{Gruber} from core AAS9/19 suggest that in comparison to the early Holocene, surface productivity increased during middle to Late Holocene and so did the intensity of denitrification. Increased denitrification must have also lead to a better preservation of OC during the Late Holocene. Moreover, Cr/Ti and Mo/Ti from the same core do suggest that bottom-waters were suboxic, which means that the OMZ intensity increased during this period. Oxygenated bottom-water conditions prevailed during the early Holocene and YD wherein sediments were rich in carbonate, aragonite and pteropods and low in OC and redox sensitive elements (von Rad *et al.*, 1999b). There is also a strong anticorrelation between CaCO₃ and δ^{15} N ($r^2 = 0.74$, n=36) which suggest that during the Late Holocene CaCO₃ dissolution was a result of increased OMZ intensity. This implies that denitrification and calcite preservation are intimately linked and reflect OMZ intensity variations. We also rule out that CaCO₃ content decrease during the Late Holocene was not a result of dilution by terrigenous material, by examining the Ti variations from this core which is a proxy for terrigenous dilution. The decrease in Ti concentrations

through the Holocene, observed in AAS9/19 is reliable as such a decreasing trend was seen in another core from the EAS, 3268G5. Higher Ti during the early Holocene was probably due to increased dust supply because of stronger winds (Agnihotri *et al.*, 2003).

OMZs have been also termed as a Carbon Maximum Zones (CMZ) since they have high dissolved inorganic carbon (DIC) concentrations and are local sources of CO₂ (Paulmier *et al.*, 2011). They could therefore induce a positive feedback for atmospheric CO₂ (Paulmier *et al.*, 2011) and lead to dissolution of CaCO₃ on the seafloor. Aragonite is a metastable polymorph of CaCO₃ and more soluble in water than calcite (Morse *et al.*, 1980). Low aragonite content and absence of pteropods during the Late Holocene from a nearby core SK 17 resulted from an increase in OMZ intensity (Fig. 1, 2k and 2l; Singh *et al.*, 2007). It is also observed in Core SK 17 that Aragonite content was high during YD (Singh *et al.*, 2007) due to inflow of oxygenated Subantarctic Mode and Antarctic Intermediate Water (SAMW-AAIW) (Böning and Bard, 2009). However the YD event is not well demarcated in all the proxies as the response of Arabian Sea to YD event is thought to be weak (Gupta *et al.*, 2011)

Shell weights of G. ruber from AAS9/19 were heavier during the early Holocene and highest during the YD probably due to the inflow of SAMW-AAIW. Lower shell weights during the Late Holocene could be a result of dissolution or due to increasing atmospheric CO₂ concentrations during the Holocene thus reducing the surface-water carbonate ion (CO₃⁼) concentrations and hence leads to thinner foraminiferal shells (Naik et al., 2010). In order to corroborate calcite dissolution during the Late Holocene, we acquired SEM micrographs of G. ruber shells at 3.57 kyr, representative of Late Holocene and 9.9 kyr, representative of early Holocene period (Fig. 4a and b). Marked dissolution features were seen on the 3.57 kyr shells, thus providing evidence that shell weight variations are due to dissolution which occurred during the Late Holocene as a result of sub-oxic bottom waters. Based on measurements of gases trapped in Antarctic ice, Indermühle et al. (1999) demonstrate that the CO₂ content of the atmosphere rose by 20 ppm between 8000 BP and the onset of the Little Ice Age. This CO₂ rise is thought to be resultant form different processes such as deposition of shallow water carbonate, carbonate compensation of land uptake during the glacial termination, terrestrial carbon uptake and release during the Holocene, the response of the ocean-sediment system to marine changes during the termination, etc. (Menviel and Joos, 2012). The CO₂ release from calcite dissolution in OMZ regions could be an important process contributing to the Holocene CO₂ rise and further studies should be aimed at quantification of CO₂ released from major OMZ regions of the world through the Holocene.

Conclusions

Productivity proxies such as OC, $\delta^{13}C_{org}$, and $\delta^{13}C_{G.\ ruber}$ in the eastern Arabian Sea Core AAS9/19 reveal that productivity was higher during the Late Holocene (~7 to 0 kyr) in comparison to early Holocene. Downcore variations in $\delta^{15}N$ of sedimentary organic matter appear to be controlled by the extent of surface productivity. Temporal variation in $\delta^{15}N$ demonstrates that the intensity of subsurface denitrification has increased from ~7 kyr BP, in response to increasing surface productivity during this period. Redox sensitive elements such as Molybdenum and Chromium indicate that the OMZ intensity also increased simultaneously suggesting a strong coupling between productivity and OMZ intensity in the eastern Arabian Sea. The increase in OMZ intensity further lead to an increase in calcite dissolution as seen in the low CaCO₃ content, lower shell weights of planktonic foraminiferal species *G. ruber* and visible dissolution features on *G. ruber* shells, during the Late Holocene. This calcite dissolution is probably a result of an increase in DIC in waters of the OMZ releasing CO₂ to the atmosphere most likely contributing to the Holocene CO₂ rise.

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Table captions:

Table 1: Radiocarbon ages and calibrated ages for Core AAS9/19 from the eastern Arabian Sea.

Figure captions:

Fig. 1.Composite SeaWiFS chlorophyll-a concentrations (mg/m³) image (source: http://gdata1.sci.gsfc.nasa.gov/), for south-west monsoon (June-September; 2008) in the Arabian Sea. A distinct difference between the western and eastern Arabian Sea is noticed with the western side showing higher chlorophyll concentrations and hence more productive. Likewise the west coast of India shows higher chlorophyll concentrations. Thick circles indicate core locations of Core AAS9/19 (this study), 3268G5 and SK17.

Fig. 2.Downcore variations from the Core AAS9/19 in, a) CaCO₃, % b) Organic Carbon %, c) δ^{15} N (‰), d) Nitrogen %, e) C/N ratios, f) δ^{13} Corg (‰), g) δ^{13} C (‰) of *G. ruber* (values are in greyscale and the black line denotes 5 point adjacent averaging), h) Ti (ppm), i) Cr/ Ti, j) Mo/Ti ratios, k) pteropod numbers, l) aragonite content, and m) shell weights of *G. ruber*.

Fig. 3.Plot ofδ¹³C (‰) versus C/N ratios to ascertain provenance of organic matter.

Fig. 4a.SEM micrographs of *G. ruber* shells at a magnification of x 700 at 3.57 kyr interval **Fig. 4b.**SEM micrographs of *G. ruber* shells at a magnification of x 700 at 9.9 kyr interval. The entire foraminifera surface displays peeling and dissolution of spines (shown by arrows).

Table 1:

Depth	Measured age (years BP)	Error (years)	Calibrated age (years BP)
(cm)			
0	1680	± 25	1115
140	7340	± 40	7678
215	10795	± 45	12035
275	10960	± 45	12335

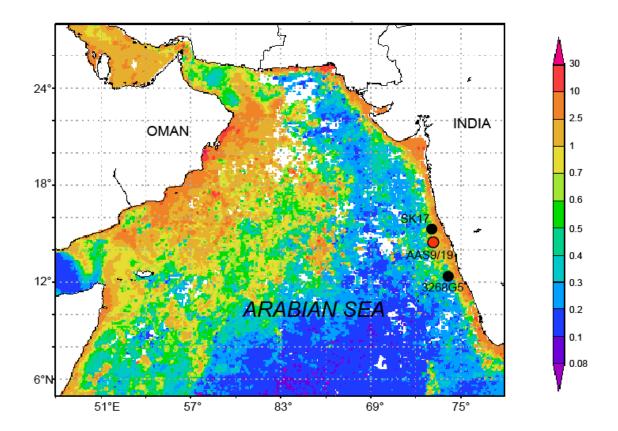


Fig. 1

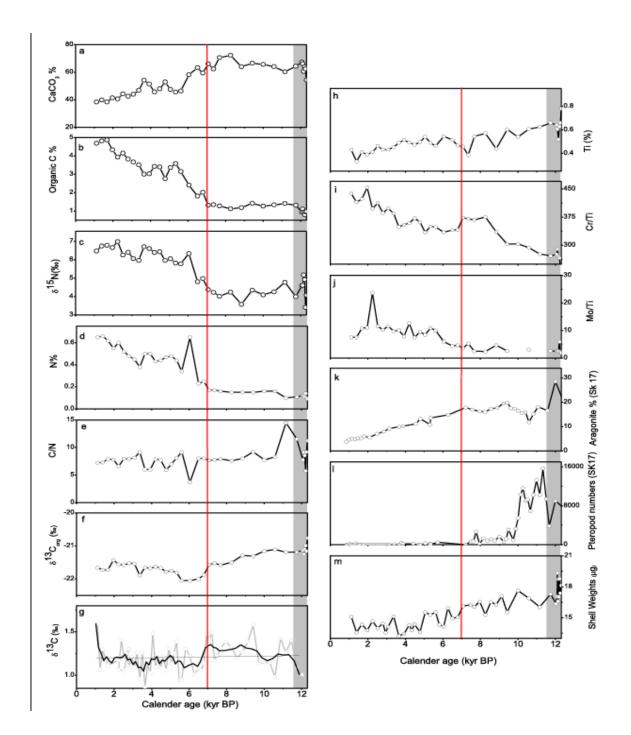


Fig. 2

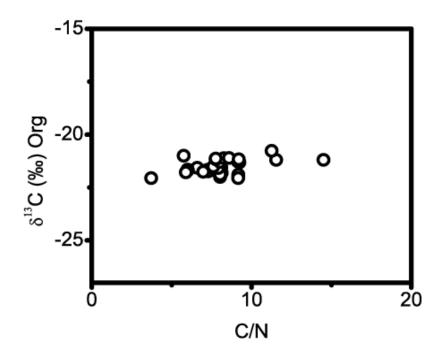


Fig. 3

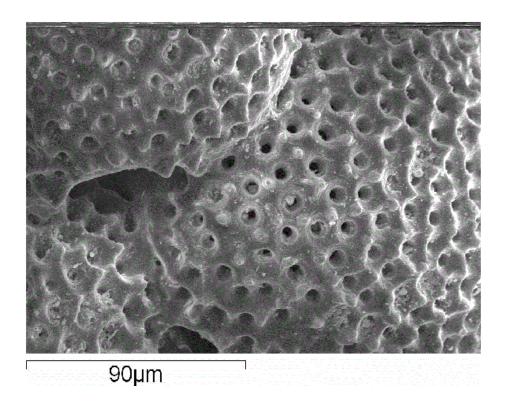


Fig. 4a

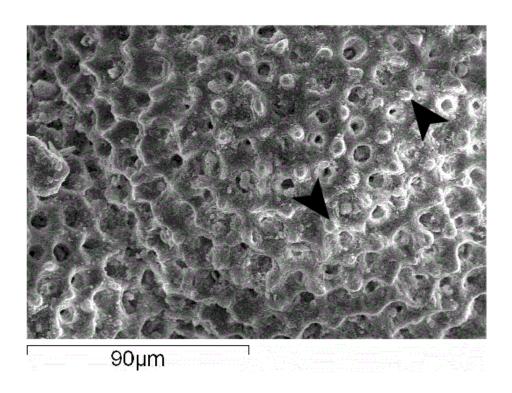


Fig. 4b